

Electronic Supplementary Materials

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Co-removal potential of heavy metals and dyes from wastewater by simultaneous adsorption with biomass residue formed from microbial treatment of lacquer residue

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Table S1 Solid-liquid ratio versus dosage

Dosage (g)	Solid-liquid ratio
0.05	1:300
0.10	1:150
0.20	1:75
0.30	1:50
0.40	1:37.5

Table S2 Concentration combinations of Pb²⁺-MB/CR composite systems

The initial concentration of MB or CR (mg/L)	The initial concentration of Pb ²⁺ (mg/L)		
	50	100	150
50	(50, 50)	(50, 100)	(50, 150)
100	(100, 50)	(100, 100)	(100, 150)
150	(150, 50)	(150, 100)	(150, 150)

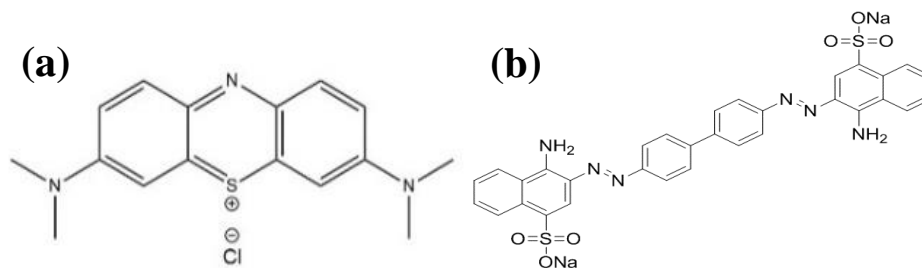


Fig. S1 The structural formulas of MB and CR

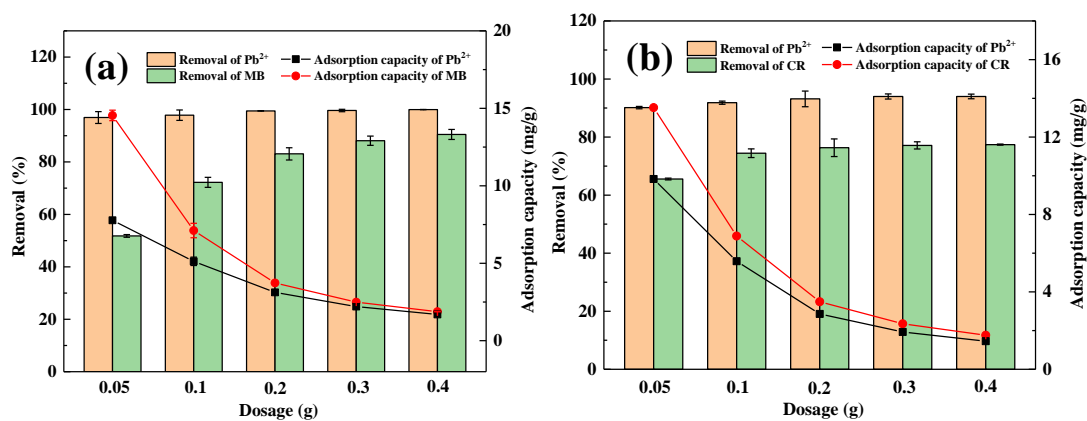


Fig. S2 Effect of dosage of LBM on adsorption of Pb²⁺-MB (a) and Pb²⁺-CR(b) composite systems

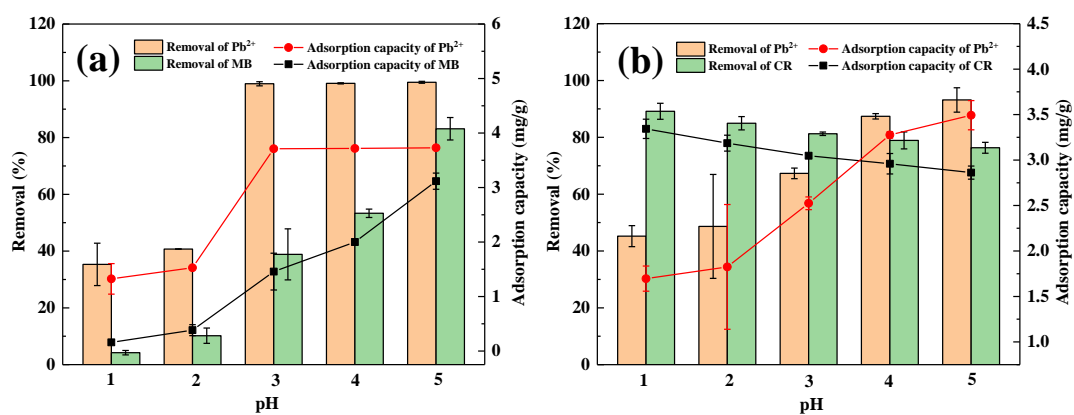


Fig. S3 Effect of system pH on adsorption of Pb²⁺-MB (a) and Pb²⁺-CR(b) composite systems

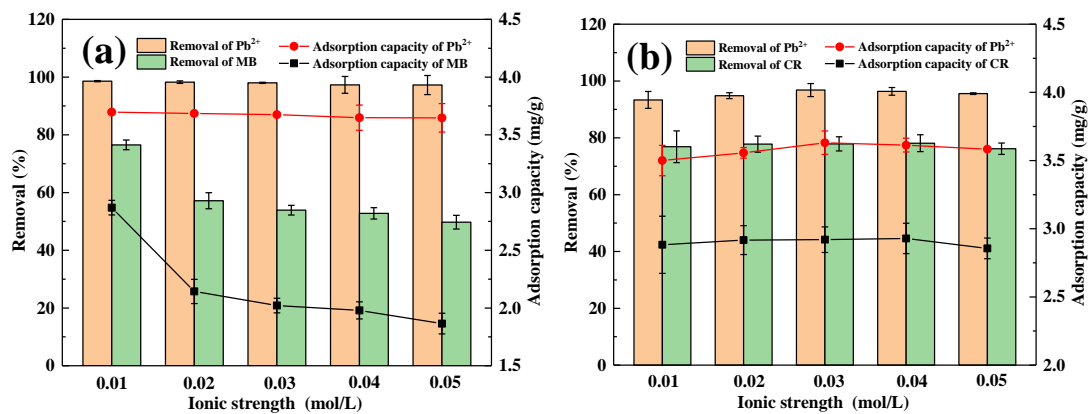


Fig. S4 Effect of ionic strength on adsorption of Pb²⁺-MB (a) and Pb²⁺-CR(b) composite systems

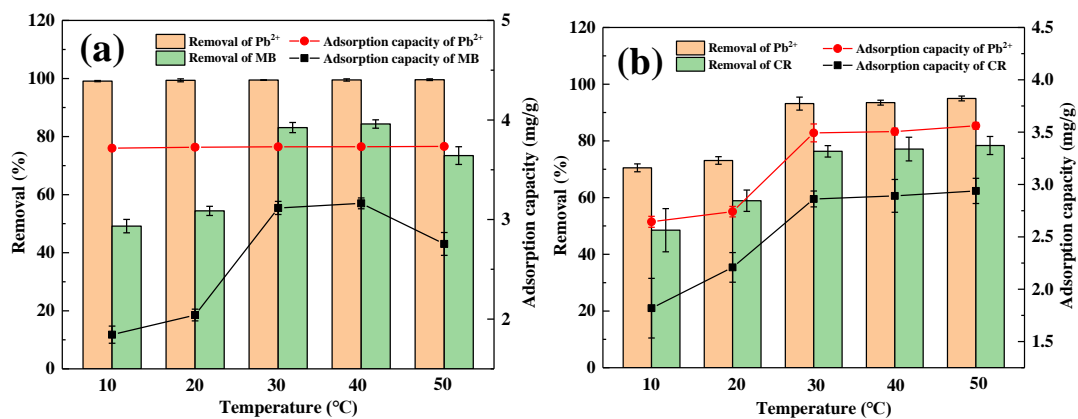


Fig. S5 Effect of temperature on adsorption of Pb²⁺-MB (a) and Pb²⁺-CR(b) composite systems

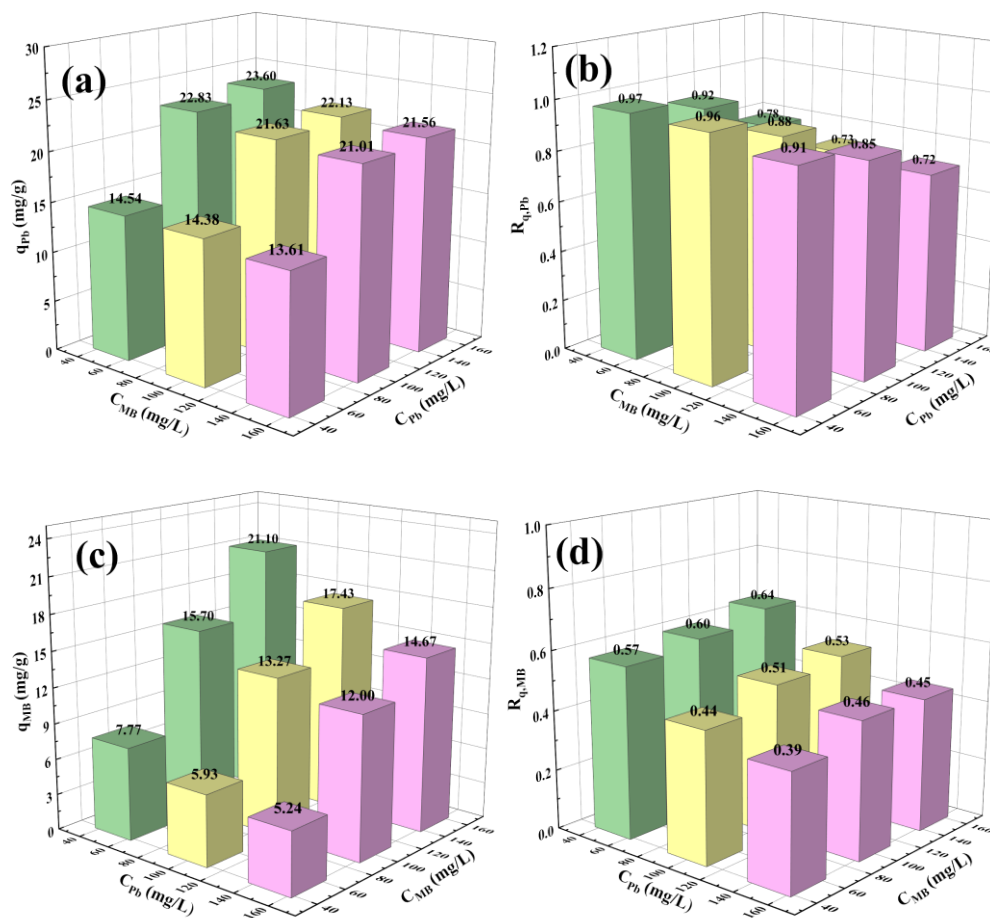


Fig. S6 Interaction between Pb^{2+} and MB in composite system, and their q (a, c) and R_q (b, d) values

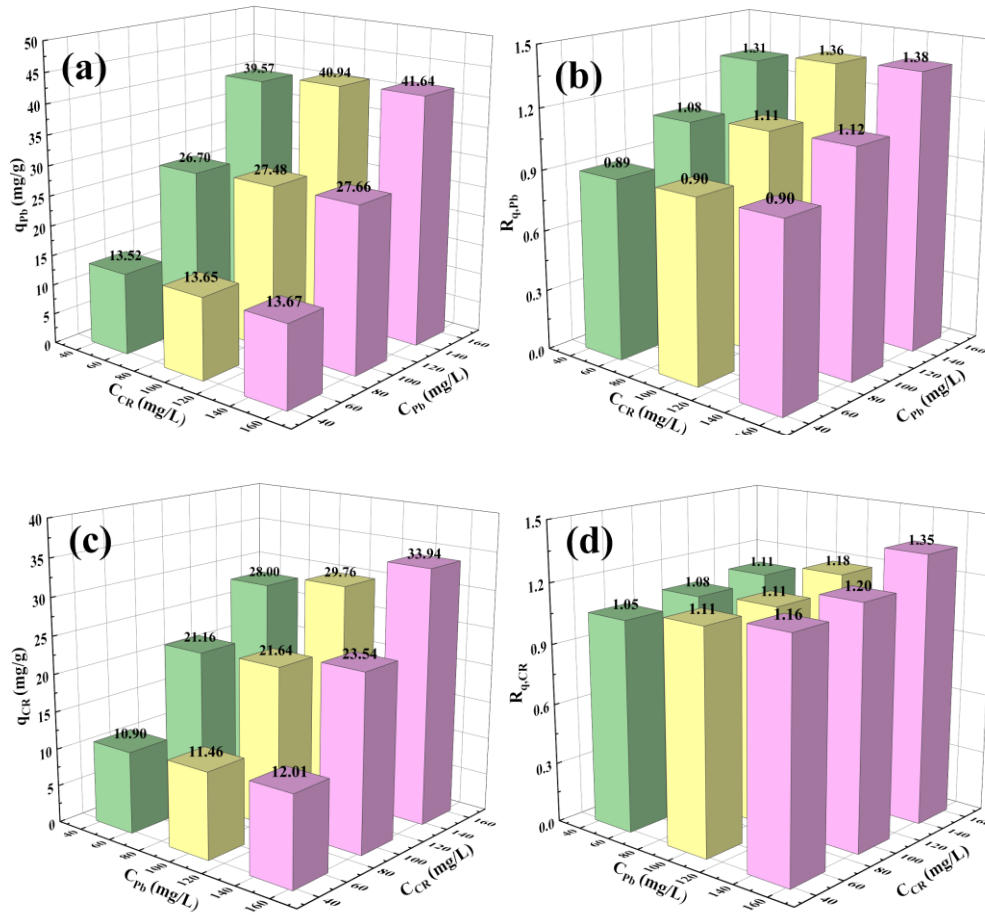


Fig. S7 Interaction between Pb^{2+} and CR in composite system, and their q (a, c) and R_q (b, d) values

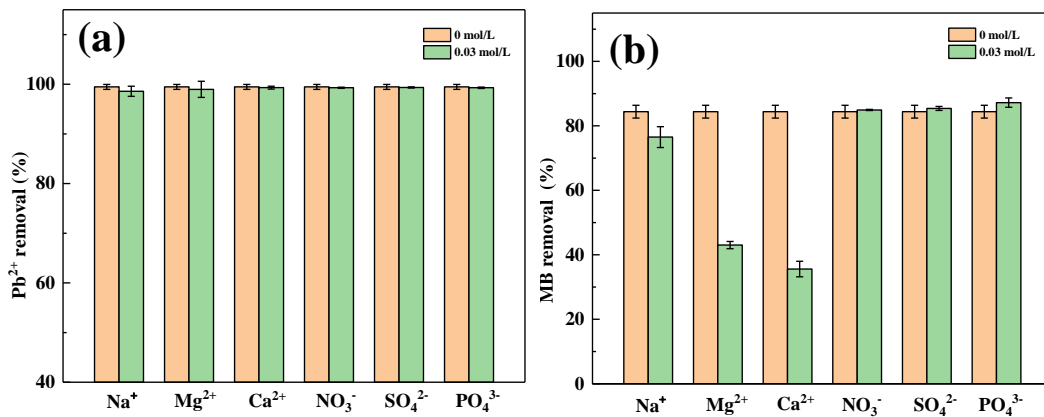


Fig. S8 Effect of coexisting cations (Na^+ , Mg^{2+} , K^+) and anions (NO_3^- , SO_4^{2-} , PO_4^{3-}) on adsorption in Pb^{2+} -MB composite system

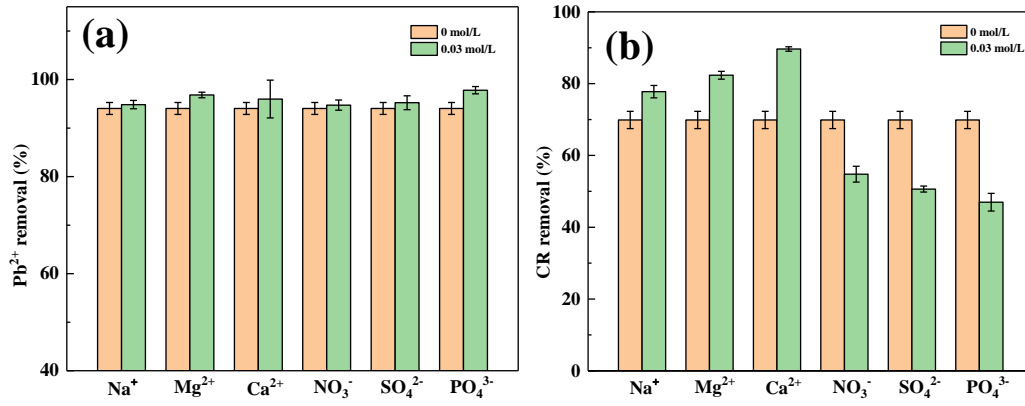


Fig. S9 Effect of coexisting cations (Na⁺, Mg²⁺, K⁺) and anions (NO₃⁻, SO₄²⁻, PO₄³⁻) on adsorption in Pb²⁺-CR composite system

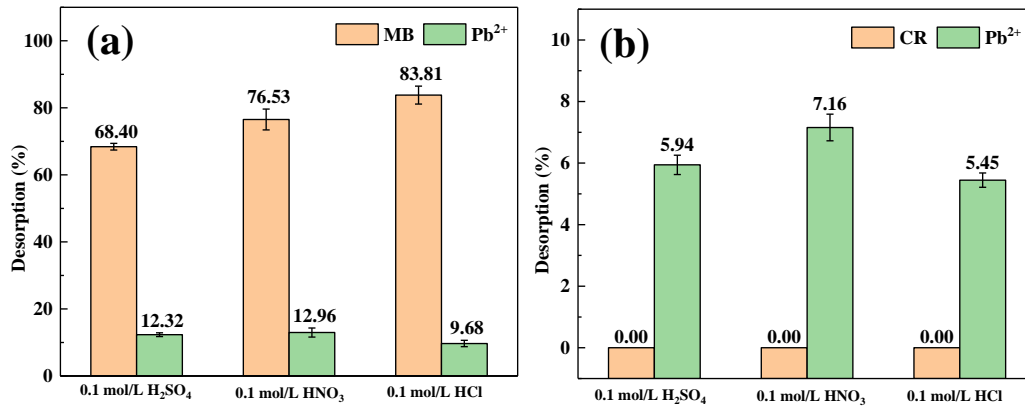


Fig. S10 Desorption efficiency of adsorbed LBM with different desorbents