



Studies on characteristics of nanostructure of N-TiO₂ thin films and photo-bactericidal action*

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Abstract: *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) from Ming lake were decomposed by photocatalytic nanostructure N-TiO₂ thin films in a photo-reactor under UV irradiation. The different thickness nanostructure N-TiO₂ thin films coated on mesh grid were prepared by sol-gel method and immobilized at 500 °C (films A) or 350 °C (films B) for 1 h in a muffle furnace. The results showed that N-TiO₂ thin film B (8.18 nm thickness, 2.760 nm height and 25.15 nm diameter) has more uniform granular nanostructure and thinner flat texture than N-TiO₂ thin film A (12.17 nm thickness, 3.578 nm height and 27.50 nm diameter). The bactericidal action of N-TiO₂ thin film A and film B for *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis varniger* strain (AS1.439) were investigated in this work. More than 95% of photocatalytic bactericidal efficiency for *Pseudomonas aeruginosa* strain (AS1.50) and 75% for *Bacillus subtilis* strain (AS1.439) were achieved by using N-TiO₂ thin films-B for 70~80 min of irradiation during the photo-bactericidal experimental process. The results indicated that the photo-induced bactericidal efficiency of N-TiO₂ thin films probably depended on the characteristics of the films.

Key words: Photocatalytic degradation, Photo-bactericidal action, N-TiO₂ thin film

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INTRODUCTION

Matsunaga *et al.* (1985) first discovered the bactericidal activity of photocatalytic TiO₂ reactions. Using TiO₂ powder particles in the bactericidal action and photocatalytic degradation for the oxidation of organic and inorganic water pollutants were extensively studied (Michael *et al.*, 1999). However, conventional TiO₂ powder catalysts were characterized by difficult separation and reproduction after reaction, so the application of titanium dioxide thin film has attracted much attention in recently years. We investigated the relationship between the antibacterial activity and the properties of the N-TiO₂ thin films, and found that the crystal structure (anatase, rutile) and morphology of particles size are important factors

affecting the photocatalytic bactericidal action. The viability of *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) during the N-TiO₂ photocatalytic reactions are also discussed.

MATERIALS AND METHODS

Preparation of nanostructure N-TiO₂ thin film

The nanostructure N-TiO₂ thin film was prepared by using the sol-gel method (Fig.1). A certain amount of tetrabutylorthotitanate and acetylacetone were mixed together first, then ethanol was added with vigorous stirring for 1 h at room temperature. After that, distilled water and nitric acid were added under stirring into the above solution. The resultant alkoxide solution was further mixed with carbamide/chloroethylamine solution for 1 h and preserved at room temperature for full hydrolysis to form the N-TiO₂ colloidal sol.

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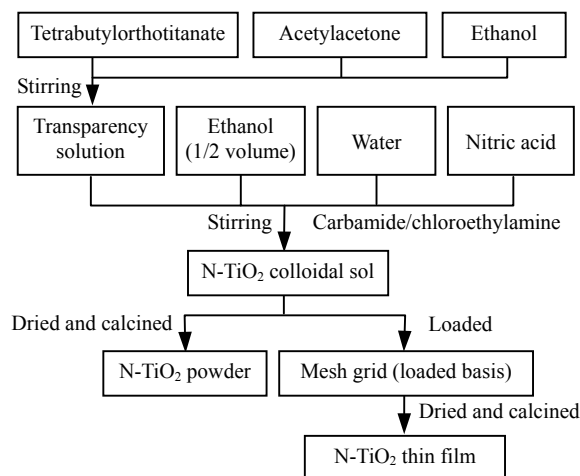


Fig.1 Preparation schematic of N-TiO₂ powder and films

The mesh grid (156.2 mm×25.4 mm×1.5 mm), which was pretreated as the loaded basis for thin films, was dipped into the N-TiO₂ colloidal sol solution for 2 min, then withdraw from the solution at rate of 1 mm/s. The film was formed on mesh grid and dried at ambient temperature for 24 h. The mesh grid coated with gel films was treated at different temperature (film A at 500 °C and film B at 350 °C/500 °C) for 1 h in a muffle furnace.

For film A, the mesh grid coated with the gel was sintered at 500 °C for 1 h with temperature raising rate of 3 °C/min in the muffle furnace, then repeatedly coated and calcined several times and finally sintered at 500 °C. For film B, the mesh grid coated with the gel films was sintered at 350 °C for 1 h, then repeatedly coated and calcined several times, with the last coating film being sintered at 500 °C for 1 h. Different thickness films were obtained by repeating the above process several times.

Characteristics of N-TiO₂ films

The surface morphology of the films and nanoparticles were determined by atomic force microscope (AFM, Dimension 3100, American DI). The crystallite size and phase of the N-TiO₂ thin films were determined by BDX3300 X-ray diffraction (XRD) measurements, carried out with Cu K_α radiation (λ=1.54 nm). The accelerating voltage and the applied current were 40 kV and 40 mA, respectively.

Setup of photocatalytic reactor system

Fig.2 is a schematic diagram of the experimental

apparatus. The photo-reactor with cooling jacket was assembled with UV light (mercury lamp 20 W, 30 cm long at wavelength 365 nm), covered with 3 mm thick transparent quartz glass cannula and immersed in the reaction solution at a constant temperature, used as a source of UV radiation. Sterile compressed air was bubbled into the reactor through air line. The incident UV radiation for the reactor was measured by ultra-violet luminometer.

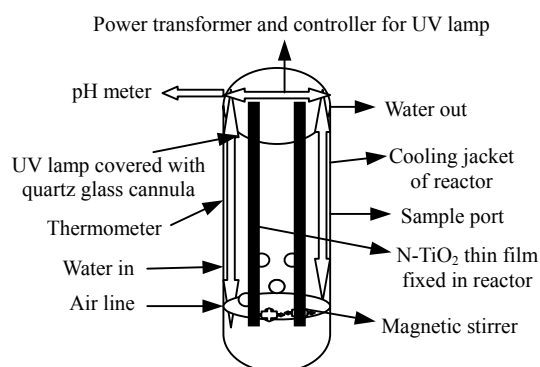


Fig.2 Diagram of photo-degradation reactor

Cultivation of microbial strain

Pseudomonas aeruginosa strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) were isolated from lake water and grown aerobically in 500 ml liquid nutrient broth at 37 °C on a rotary shaker (120 r/min) for 16 h. The medium was autoclaved at 121 °C for 30 min to ensure the sterility for testing. *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) cells were cultivated for 2 d and harvested by centrifugation at 8000×g for 10 min, washed and resuspended in the 1000 ml phosphate-buffer solution (0.2 mol/L, pH=7.0). The number of viable cells in the suspensions samples was counted by using the spread plate colony-counting method. Serial dilution of the cells was performed to obtain initial concentration of 10⁶ colonies forming units per milliliter (CFU/ml) for photocatalytic bactericidal experiment.

Photocatalytic bactericidal action procedure and cell viability assay

The 750 ml cells-suspended solution with initial concentration of 10⁶ CFU/ml in 1 L reactor was illuminated by irradiation of 20 WUV-Hg lamps (GGZ-20 W) with emitting spectral maximum at 365 nm peak wavelength. The light intensity reaching the

surface of the films fixed at the glass reactor was approximately 18 W/m^2 , which was measured by UV-meter with the peak intensity at 365 nm (model J-1221, UVP Inc. USA). N-TiO₂ film A or N-TiO₂ film B was placed into the reactor. Aliquots of cells solution were taken at 15 min intervals after irradiation with continuous stirring magnetically for uniform distribution, were added into the tube containing phosphate buffer solution for CFU testing.

The CFU were determined by plating aliquots of serially diluted suspensions on duplicate nutrient agar plates. All plates were incubated for 24 h at 30 °C, the viable number of cells were counted. The loss of cell viability was examined by the viable count procedure. All experiments were performed at least in triplicate.

RESULTS AND DISCUSSION

Characteristics of the nanostructure N-TiO₂ thin films

It is well known that the films surface morphology is dependent on the calcining treatment temperature. Different size particles on the N-TiO₂ films will be obtained at different sintering temperature for the immobilized films treatment. The film styles formed were divided into film A and film B based on a different sintering temperature. The results are shown in Fig.3.

Fig.3 is the AFM images of the N-TiO₂ thin films with five layers in film A and film B. N-TiO₂ film B has more uniform granular microstructure and

thinner flat texture than N-TiO₂ film A (Fig.3). The thickness and the average diameter of the particle size of the film were 12.17 nm and 27.50 nm for film A, 8.18 nm and 25.15 nm for film B respectively determined by AFM and XRD technique. The granular microstructure is the key factor influencing the photocatalytic activity (Jang *et al.*, 2001). In the N-TiO₂ film B, the mass fraction of crystallitic anatase present in granular nanostructure are mostly up to 94.8% higher than 65.2% of mass fraction in the N-TiO₂ film A.

Zarzycki *et al.*(1982) explained the mechanism of the particles growth of gel films in the calcining process. From Fig.4 shows that during calcining treatment for the film A process, there are many H₂O molecules and hydroxyl groups forming hydrogen bonds on the sol-gel surface of the N-TiO₂ film coated on mesh grid, when N-TiO₂ gel films were treated at 500 °C five times, hydrogen bonds between the two molecules would turn into chemical covalent bond during sintering process, then link close N-TiO₂ crystallitic particles tightly, result in the particles growing and solid cluster conglomeration, consequently, lead to the rapidly increasing number of larger diameter particles. In film B, N-TiO₂ thin film (8.18 nm thickness, 2.760 nm height and 25.15 nm average nanoparticle diameter) has more uniform granular microstructure and thinner flat texture than N-TiO₂ thin film A (12.17 nm thickness, 3.578 nm height and 27.50 nm average nanoparticle diameter) because low sintering temperature decreases the coagulation and growth rate of particles, hence, the mass

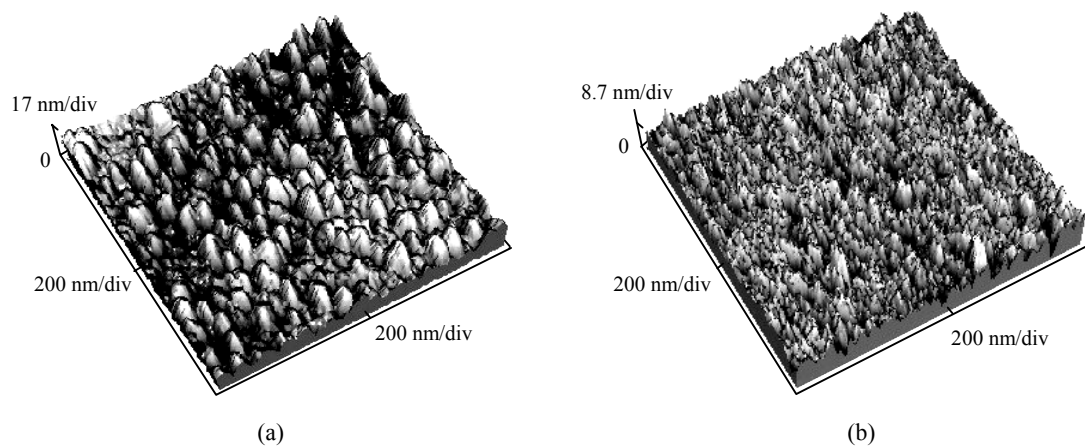


Fig.3 AFM images of the N-TiO₂ thin films. (a) Film A (at 500 °C, five layers); (b) Film B (at 350 °C/500 °C, five layers)

fraction of N-TiO₂ anatase nanostructure has more reactive center in N-TiO₂ thin film B than in N-TiO₂ thin film A, which is probably due to the smaller size of anatase nanoparticle.

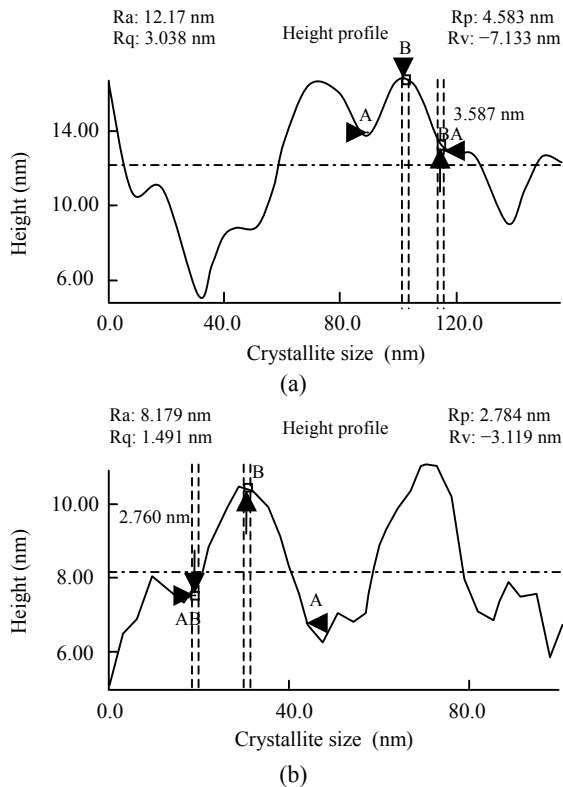


Fig.4 The size of nanoparticle on N-TiO₂ thin film for (a) film A (at 500 °C, five layers) and (b) film B (at 350 °C/500 °C, five layers)

Effect of N-TiO₂ nanoparticles of thin film on photocatalytic bactericidal activity

Pseudomonas aeruginosa strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) cells inactivation was found to be rapid in the presence of nanostructured N-TiO₂ films during UV irradiation.

Fig.5 shows the effect of illuminated N-TiO₂ thin films B (8.18 nm thickness) on the sterilization of *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis* strain (AS1.439) in comparison with N-TiO₂ thin films A (12.17 nm thickness) under the conditions of initial concentration of 10⁶ CFU/ml, 18 W/m² of UV intensity. The viability of cells is determined by colonies counting. The survival curves are shown in Fig.5 for *Pseudomonas aeruginosa* strain (AS1.50) and *Bacillus subtilis* strain (AS1.439), respectively.

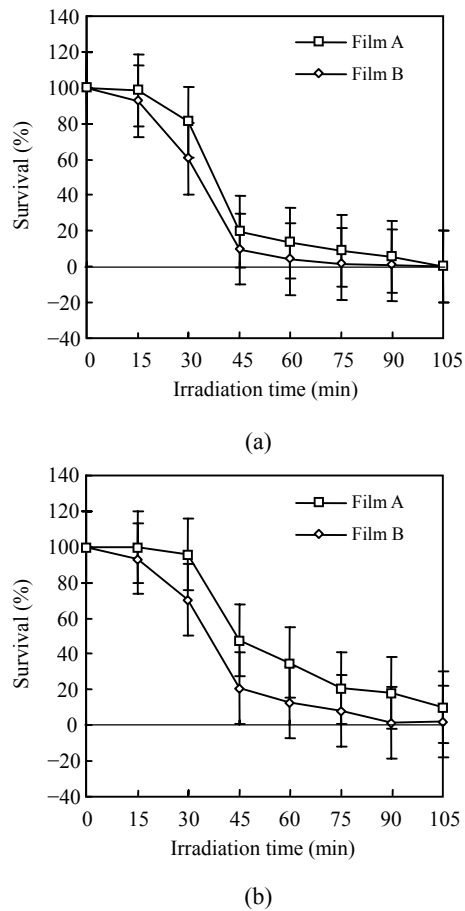


Fig.5 Effect of the N-TiO₂ films on the viability of (a) *Pseudomonas aeruginosa* strain (AS1.50) and (b) *Bacillus subtilis* strain (AS1.43)

More than 95% photocatalytic sterilization rate for *Pseudomonas aeruginosa* strain (AS1.50) and 75% for *Bacillus subtilis* strain (AS1.43) were achieved by using N-TiO₂ thin films B for 70~80 min of irradiation during the photo-bactericidal experimental process. The results indicated that the anti-bactericidal effect of N-TiO₂ film was associated directly with the thickness of films (Trapalis et al., 2003). The photo-bactericidal action no *Bacillus subtilis* strain (AS1.439) and *Pseudomonas aeruginosa* strain (AS1.50) treated by N-TiO₂ thin films B was shown to be more effectively than N-TiO₂ thin films A. *Bacillus subtilis* strain (AS1.439) showed stronger resistance to photocatalytic sterilization than the *Pseudomonas aeruginosa* strain (AS1.50) (Fig.5).

The photocatalytic chemical reaction mainly occurs on the surface of the films, indicating that the photocatalytic reaction mainly occurs on the specific

particles size, the photocatalytic activity is enhanced with decreasing nanoparticles diameter and increasing contact area between the photocatalyst and target material (Maness *et al.*, 1999). In addition, the higher temperature treatment could not only cause the aggregation of N-TiO₂ particles, but also partially change crystalline phase from the anatase to rutile which probably result in losing some active sites of the films for photocatalytic reaction during different thermal treatment (Trapalis *et al.*, 2003). The experimental result suggested photo-induced bactericidal efficiency of N-TiO₂ was affected by the morphology and thickness of N-TiO₂ films, the grain size of nanoparticles, surface area, mass anatase composition of N-TiO₂.

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