

# In-situ investigation of melting characteristics of waste selective catalytic reduction catalysts during harmless melting treatment

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# Paper content overview

- Thermal melting technology was applied to dispose of waste SCR catalyst.
- Additive formulas were determined by multicomponent phase diagrams.
- Melting characteristics and mineral phase transformation of waste SCR catalyst with additives were investigated by means of heating stage microscope, TG/DSC, Factsage simulation and XRD.
- The leaching toxicity of heavy metal elements was tested by ICP-OES.
- The additives formula of  $\text{Fe}_2\text{O}_3\text{-CaO-SiO}_2\text{-Al}_2\text{O}_3$  demonstrated the optimal fluxing behavior.

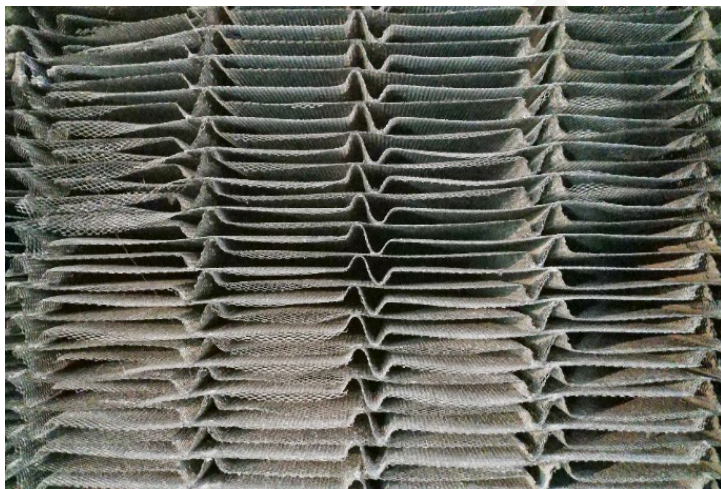
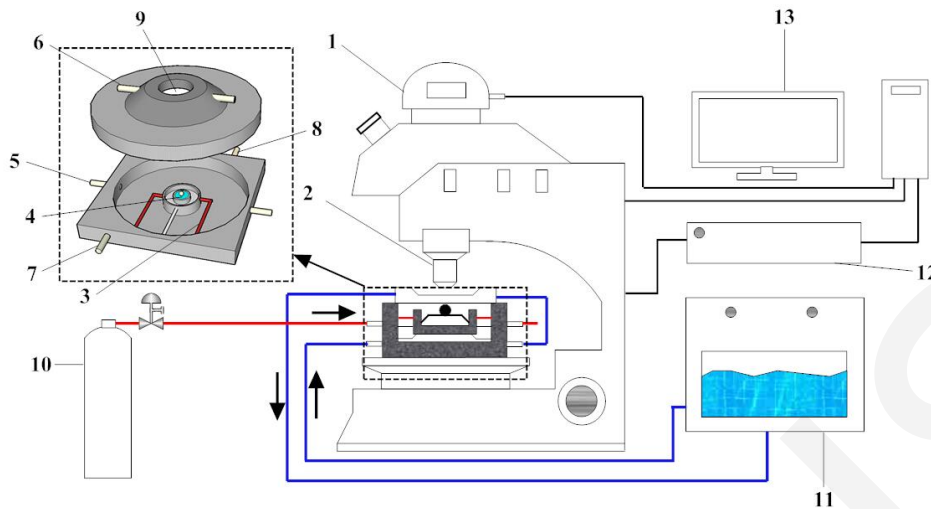


Plate type waste SCR catalyst



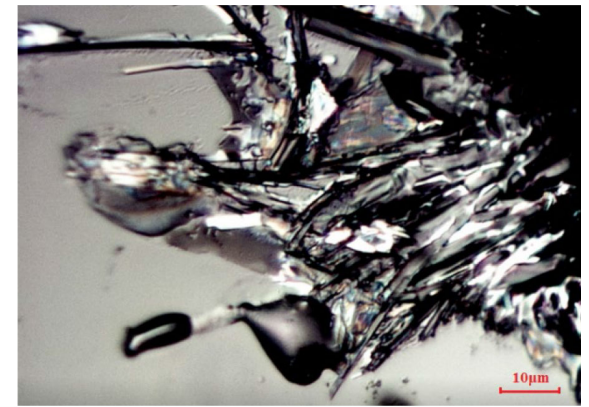
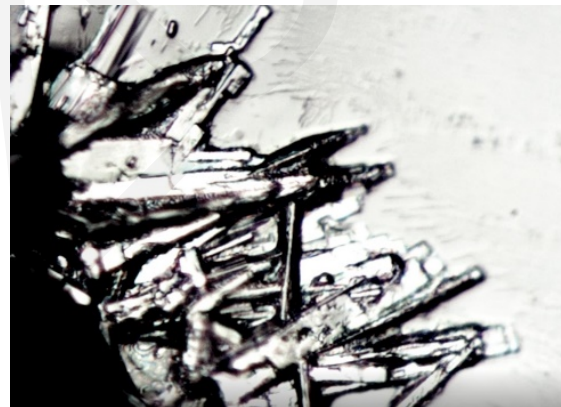
Pulverized waste SCR catalyst

# Experimental facility



1. Digital Camera;
2. Microscope Lens;
3. Heating Wire;
4. Sapphire Substrate;
5. Water Inlet;
6. Water Outlet;
7. Gas inlet;
8. Gas Outlet;
9. Quartz Glass;
10. Air Cylinder;
11. Cooling Water Tank;
12. Temperature Controller;
13. Online Computer System

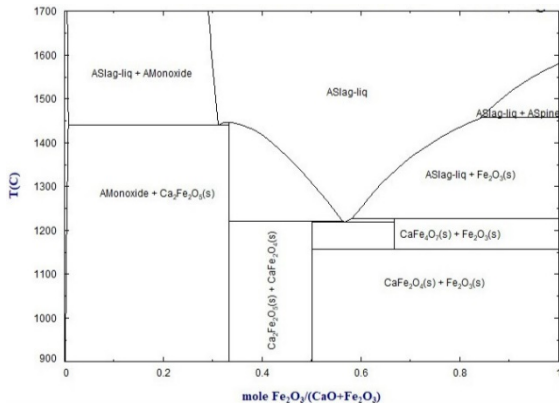
Heating stage microscope



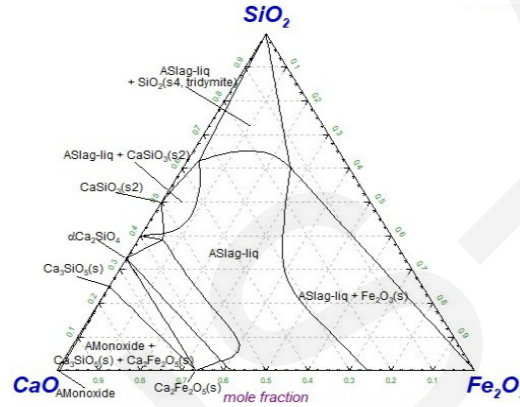
Melted samples with additives added

# Experimental facility

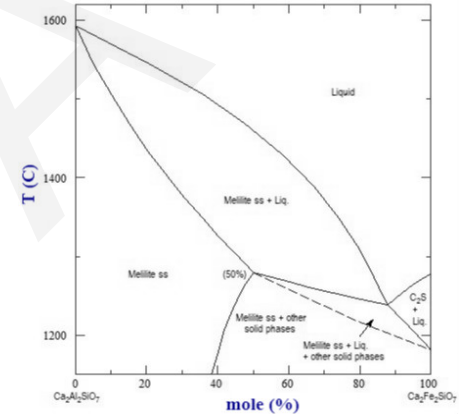
**CaO-Fe<sub>2</sub>O<sub>3</sub>**



**CaO-SiO<sub>2</sub>-Fe<sub>2</sub>O<sub>3</sub>**



**CaO-SiO<sub>2</sub>-Fe<sub>2</sub>O<sub>3</sub>-Al<sub>2</sub>O<sub>3</sub>**



**Multicomponent phase diagrams**

Samples	Catalyst	CaO	Fe <sub>2</sub> O <sub>3</sub>	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>
A1	10.00	18.00	72.00	—	—
A2	50.00	10.00	40.00	—	—
A3	90.00	2.00	8.00	—	—
B1	10.00	24.30	54.00	11.70	—
B2	50.00	13.50	30.00	6.50	—
B3	90.00	2.70	6.00	1.30	—
C1	10.00	11.70	70.20	5.94	2.16
C2	50.00	6.50	39.00	3.30	1.20
C3	90.00	1.30	7.80	0.66	0.24

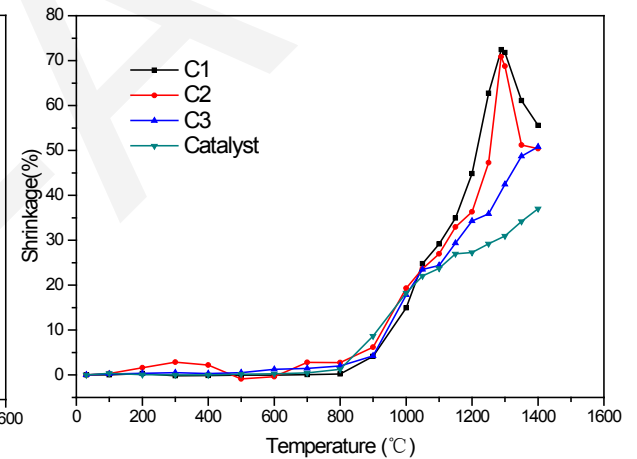
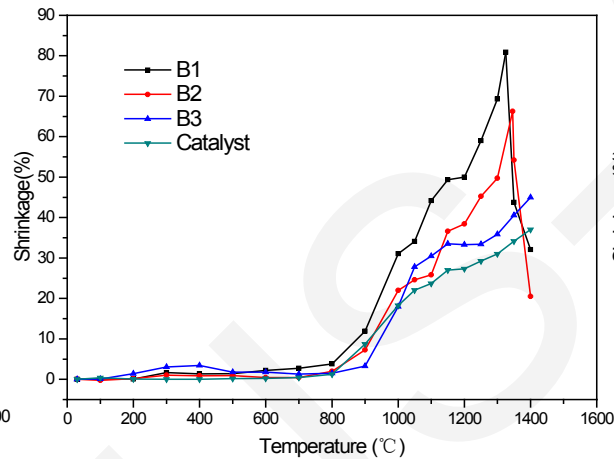
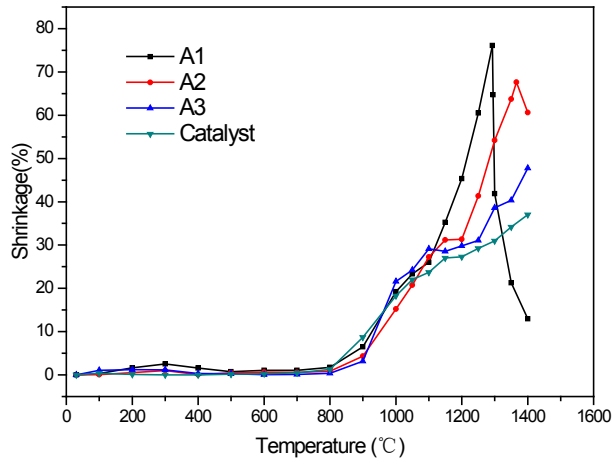
**Optimal additive formula**

**6.50 wt. % CaO**  
**3.30 wt. % SiO<sub>2</sub>**  
**39.0 wt. % Fe<sub>2</sub>O<sub>3</sub>**  
**1.20 wt. % Al<sub>2</sub>O<sub>3</sub>**

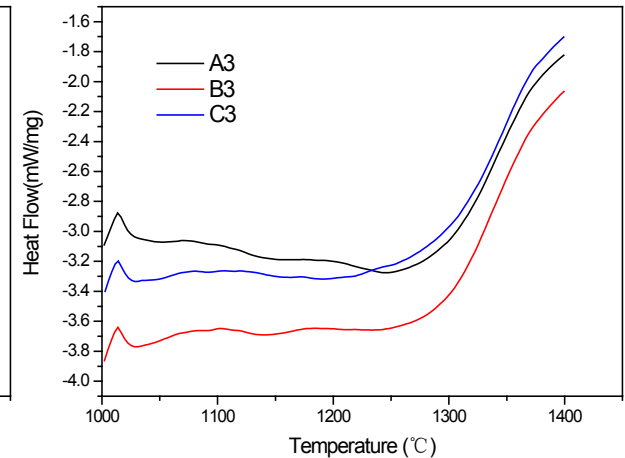
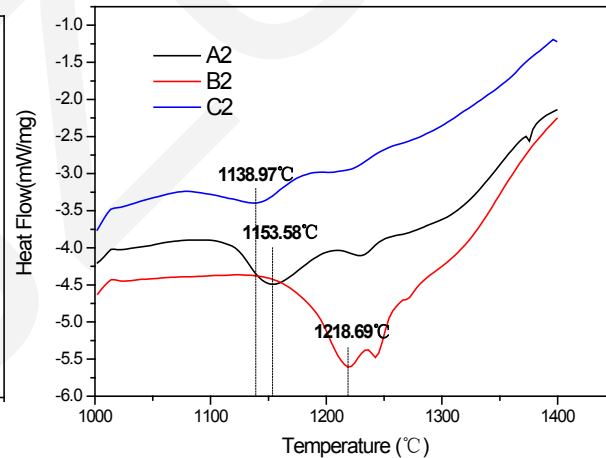
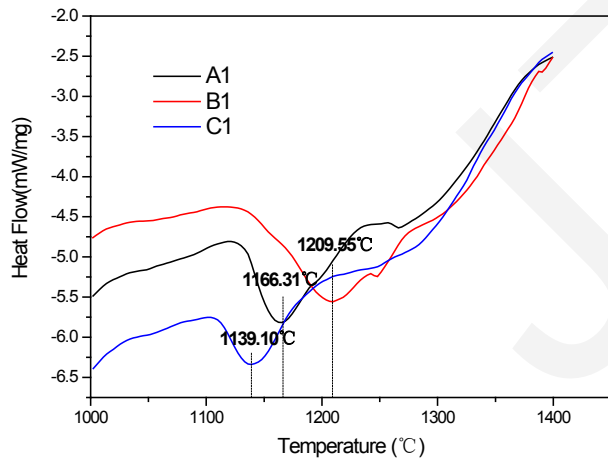


# Melting treatment results

## Effect of additives on the area shrinkage

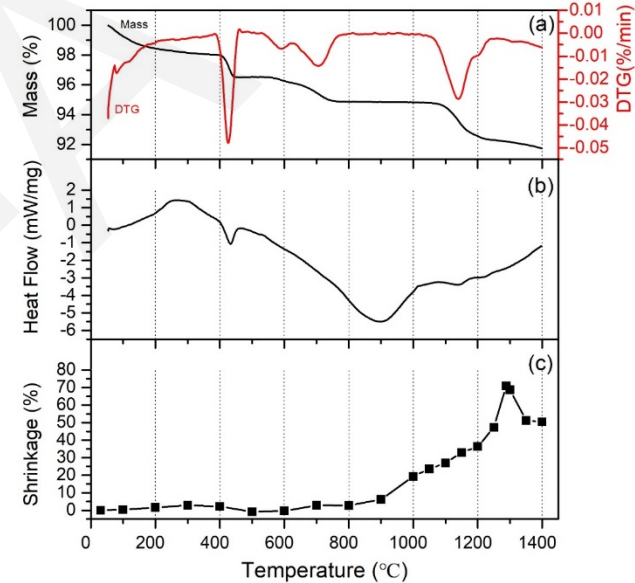
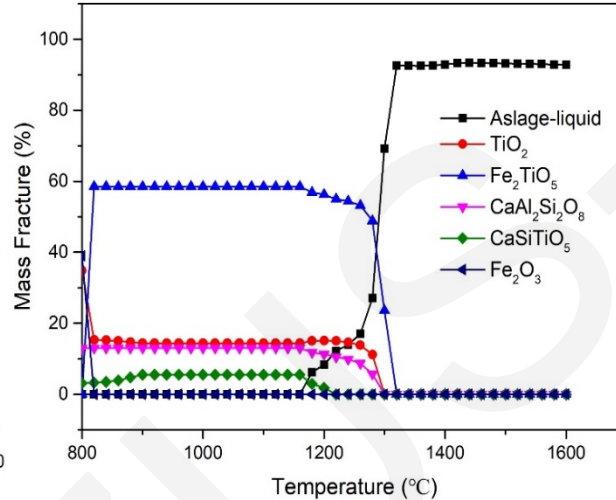
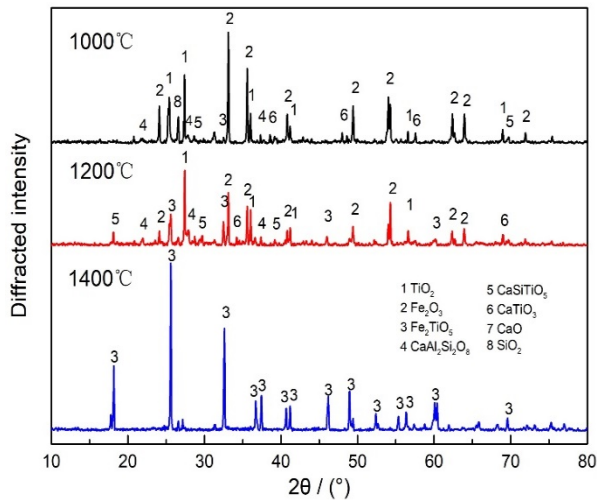


## Effect of additives on the melting temperature (DSC curves)



# Melting treatment results

## Melting characteristics and mineral phase transformation of samples with optimal additives



XRD patterns and Factsage simulation results

TG-DSC results and Area shrinkage

## Heavy metal leaching toxicity tested by ICP-OES

Elements	V	As	Pb	Se
Raw sample	10.648	1.054	0.195	0.347
Molten slag	0.178	0.025	0.048	0.003
Regulatory standard limits	0.2	5	5	1

(ppm)

# Conclusions

- The waste SCR catalyst is a  $\text{TiO}_2$ -based oxide mixture with a high melting point which can be effectively reduced with proper additives.
- The additive formula of C2, namely 39.00 wt. %  $\text{Fe}_2\text{O}_3$ , 6.50 wt. %  $\text{CaO}$ , 3.30 wt. %  $\text{SiO}_2$ , and 1.20 wt. %  $\text{Al}_2\text{O}_3$  has the optimal fluxing behavior, decreasing the initial melting temperature from 1223 °C to 1169 °C.
- The melting process includes three stages: the solid reaction stage, the sintering stage, and the primary melting stage.
- Liquid phase generated from the waste catalyst flows, wets, and spreads to the surfaces of  $\text{TiO}_2$  particles and reacts with them to generate  $\text{CaTiO}_3$  and  $\text{Fe}_2\text{TiO}_5$ . The later mineral phases form low-temperature eutectics, encapsulating  $\text{TiO}_2$  and promoting the melting process.
- Leaching concentrations of heavy metals of samples with additives added after melting treatment were much lower than those of the raw waste catalyst, which demonstrated that such a method has a strong immobilization capacity for heavy metals in waste SCR catalysts.