



Bonding heterostructure mediated “photo-thermo-electric” implant: NIR-II photothermal and thermoelectric therapy for bone tumor defects

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Abstract

Recurrence of solid tumors after surgical resection is a major barrier to tissue regeneration. As an emerging treatment strategy, photo-thermo-electric therapy ablates tumor cells via photothermal effects and generates reactive oxygen species (ROS) via thermoelectric effects to disrupt heat shock proteins, thereby suppressing their protective function in tumor cells. However, conventional materials suffer from low thermoelectric efficiency and weak tissue penetration ability. In this study, we fabricated iodine-doped bismuth sulfide (I-Bi₂S₃) nanorods with bonding heterostructures to improve thermoelectric performance. The approach employed iodine doping to introduce additional electrons, thereby regulating the band structure of Bi₂S₃ and exploiting the dual low-energy vibration effect of the heterostructures to reduce thermal conductivity. More importantly, controlling the type of heterostructure modulated the bandgap width, thereby expanding the light absorption range to the higher-penetration near-infrared (NIR)-II region for deep tissue treatment. The I-Bi₂S₃ nanorods were incorporated into poly-L-lactic acid (PLLA) scaffolds to confer antitumor functionality. According to the results, the bonding heterostructures enhanced the conductivity of Bi₂S₃ and reduced its thermal conductivity, significantly enhancing thermoelectric efficacy. The heterostructures reduced the bandgap of Bi₂S₃ from 1.23 to 0.88 eV, enabling optical absorption in the NIR-II region. The ROS tests showed that the PLLA/I-Bi₂S₃ scaffold exhibited good photothermal effects and ROS generation under 1064-nm laser irradiation. The antitumor efficacy of the PLLA/I-Bi₂S₃ scaffold reached 84.6% against MG-63 cells, demonstrating its exceptional potential in cancer treatment.

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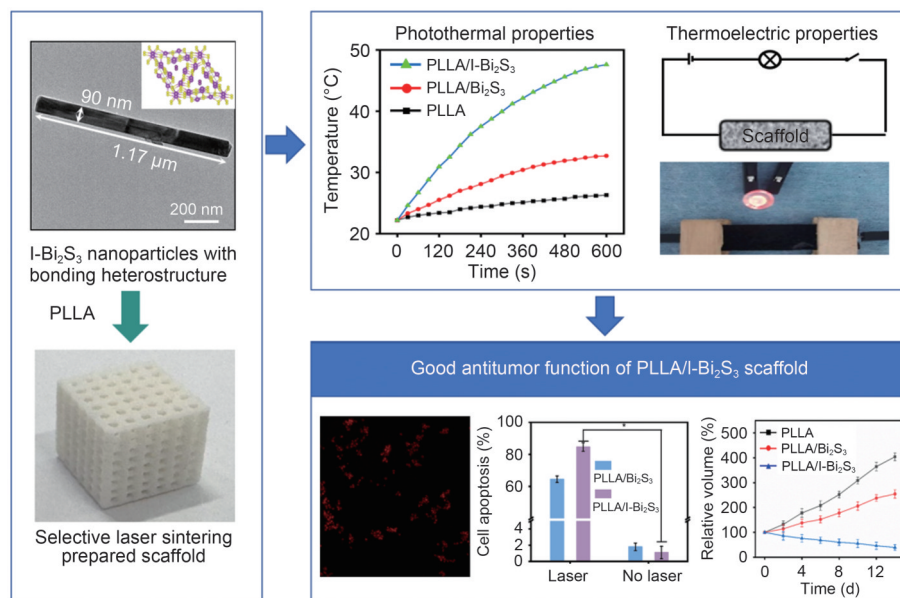
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Graphical abstract



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1 Introduction

Tumor-induced tissue defects have long posed a major challenge for the medical community, particularly osteosarcoma, which primarily occurs in children and adolescents [1, 2]. On the one hand, surgical resection rarely removes malignant tumor tissue completely; thus, recurrent tumor cells quickly invade healthy bone tissue and even newly formed bone tissue [3, 4]. On the other hand, the defective tissue after surgical resection is difficult to repair by self-healing [5]. Recently, the development of degradable implants with antitumor functions using three-dimensional (3D) printing has emerged as a promising treatment approach [6, 7]. Specifically, 3D-printed biodegradable scaffolds (such as poly-L-lactide (PLLA) and poly(lactic-co-glycolic acid)) not only precisely match the host bone defect to realize sufficient mechanical support but also serve as growth sites to promote cell adhesion and growth [8–10]. Importantly, scaffolds with antitumor functions can prevent tumor recurrence and provide a suitable growth environment for tissue regeneration [11].

Currently, the primary methods for endowing implants with antitumor capabilities include chemotherapy, photothermal therapy (PTT), and photodynamic therapy (PDT) [12]. Chemotherapy typically faces limitations due to drug resistance, low treatment efficacy, and side effects [13]. In

contrast, PTT and PDT offer benefits such as noninvasiveness, broad therapeutic ranges, and low resistance [14, 15]. However, the overexpression of heat shock proteins (HSPs) in tumor cells can diminish PTT effectiveness, and the reactive oxygen species (ROS) generated by PDT have a short duration of action [16, 17]. Photo-thermo-electric therapy has recently emerged as a promising strategy for cancer treatment, combining the strengths of PTT and PDT [18]. In this approach, PTT is first used to generate localized high temperatures to ablate tumor cells, while pyroelectric therapy leverages the temperature differences to produce ROS, which disrupt HSPs, thereby enhancing antitumor efficacy.

Bismuth sulfide (Bi₂S₃), a typical photo-thermo-electric material, has attracted considerable attention for its good biocompatibility, narrow bandgap, and good light absorption [19]. However, its low thermoelectric efficiency at normal temperatures and weak tissue penetration restrict its application in bone tissue [20]. Research has shown that inserting a third phase into binary compounds can significantly modify the electronic and phonon structures, thereby enhancing their thermoelectric performance. Constructing bonding heterogeneity to improve the overall performance is an effective strategy [21]. For example, introducing halogen elements (such as iodine, chlorine, and bromine) into Bi₂S₃ is expected to form bonding heterostructures. Iodine

exhibits high electronegativity and strong polarizability; thus, the doping of iodine (I) atoms into the Bi_2S_3 structure will partially substitute sulfur atoms and modulate the coordination environment of Bi via ionic polarization effects, thereby inducing the formation of Bi^{2+} – Bi^{2+} covalent bonds [22, 23]. In this case, covalent bonds such as Bi–S, Bi–I, and Bi–Bi rearrange within the lattice to form a bonded heterogeneous crystal cell structure (e.g., $\text{Bi}_{13}\text{S}_{18}\text{I}_2$ or $\text{Bi}_{19}\text{S}_{27}\text{I}_3$). The additional electrons introduced by iodine doping can regulate the band structure, thereby increasing the carrier concentration and enhancing electrical conductivity [24, 25]. In addition, the iodine atoms and subvalent dimers introduce dual low-energy vibrations, which reduce lattice thermal conductivity, thereby improving thermoelectric conversion efficiency. Moreover, the bonded heterogeneous structure forms with good interfacial compatibility, avoiding the interface resistance caused by lattice mismatch and further enhancing thermoelectric stability.

More importantly, this process also holds promise for extending the light absorption range of Bi_2S_3 into the near-infrared (NIR)-II region with deep penetration, thereby enabling photothermal responsiveness in deep bone tissue environments (>3 cm). This is because iodine doping and the formation of bonded heterogeneous structures reconstruct the Bi_2S_3 band structure, introducing new donor levels between the valence band (VB) and conduction band (CB) and significantly narrowing the bandgap. If the doping process can be precisely tuned to achieve a bandgap width within the range of 0.73–1.24 eV, electrons can be excited by lower-energy photons corresponding to the NIR-II region, enabling photothermal conversion under NIR-II irradiation [26, 27]. Additionally, the presence of multiple bonding states within the heterostructure may cause local polarity variations, thereby generating weak built-in electric fields within the material. These fields facilitate the separation of photogenerated electron–hole pairs, further improving photothermal conversion efficiency.

In this study, iodine-doped Bi_2S_3 (I- Bi_2S_3) nanoparticles (NPs) with $\text{Bi}_{13}\text{S}_{18}\text{I}_2$ and $\text{Bi}_{19}\text{S}_{27}\text{I}_3$ heterostructures were first synthesized using a hydrothermal reaction method to fabricate a functional NP with enhanced photothermal–thermoelectric properties. The resulting I- Bi_2S_3 NPs were then incorporated into PLLA scaffolds using laser 3D printing technology for antitumor therapy. The morphology and phase composition of the synthesized I- Bi_2S_3 NPs were comprehensively analyzed. The functional characteristics of the PLLA/I- Bi_2S_3 scaffolds were investigated, including photothermal and thermoelectric performance and the corresponding ROS generation ability. The antitumor function of the scaffold was detected through cell and animal experiments. Furthermore, the antitumor mechanism was studied in detail. This study proposes an antitumor scaffold based on photothermal–thermoelectric therapy, providing a

therapeutic approach for tissue regeneration following osteosarcoma resection.

2 Experimental section

2.1 Material source and synthesis

2.1.1 Material source

PLLA powder ($M_w=1.5\times10^4$ Da, melting point of approximately 175 °C) was obtained from Polymtek Biomaterial Co., Ltd. (Shenzhen, China). Bismuth nitrate pentahydrate ($\text{Bi}(\text{NO}_3)_3\cdot5\text{H}_2\text{O}$, 99%), thiourea ($\text{CS}(\text{NH}_2)_2$, 99%), and potassium iodide (KI, 99%) were purchased from Aladdin Biological Co., Ltd. (Shanghai, China).

2.1.2 Synthesis of I- Bi_2S_3

I- Bi_2S_3 was synthesized using a hydrothermal reaction method. In detail, 0.1 mmol/L of $\text{Bi}(\text{NO}_3)_3\cdot5\text{H}_2\text{O}$ was dissolved in 30 mL of deionized water and stirred for 2 h to ensure complete dissolution. Next, 38 mg of $\text{CS}(\text{NH}_2)_2$ and 0.2 mg of KI were added, and the mixture was stirred for an additional 30 min. The resulting solution was transferred into a 50-mL Teflon-lined autoclave and heated at 200 °C for 8 h. After cooling, the black precipitate (I- Bi_2S_3 powder) was washed, centrifuged with deionized water, and dried in a vacuum oven. For comparison, Bi_2S_3 powder was synthesized under identical conditions, excluding the addition of KI.

2.2 Characterization

The micro-morphology of the I- Bi_2S_3 NPs was observed using transmission electron microscopy (TEM; Tecnai F20, FEI, USA). The elemental composition of the I- Bi_2S_3 NPs was examined using energy-dispersive X-ray spectroscopy (EDS). The phase structure and surface chemical composition were investigated using X-ray diffraction (XRD; Empyrean, the Netherlands) and X-ray photoelectron spectroscopy (XPS; Thermo-VG Scientific Ltd., UK), respectively. Electrochemical properties were evaluated using an electrochemical workstation (CHI660E, China) with a standard three-electrode setup, where Ag/AgCl, platinum foil, and the sample served as the reference, counter, and working electrodes, respectively. Electrochemical impedance spectroscopy (EIS) and Mott–Schottky analyses were performed using a 2.5-mmol/L potassium ferricyanide solution, and photocurrent measurements were performed using a 0.5-mol/L Na_2SO_4 solution. EIS data were collected across a frequency range of 100 kHz–0.1 Hz with an amplitude of 0.5 V [28, 29]. Ultraviolet–visible (UV–Vis) absorption and diffuse reflectance spectra were recorded using a UV-2700 spectrophotometer (Shimadzu, China).

To elucidate the changes in the energy band structure and density of states (DOS) after the iodine doping of Bi_2S_3 , first-principles calculations were performed using the Cambridge Sequential Total Energy Package simulation program based on density functional theory (DFT). The Perdew–Burke–Ernzerhof exchange–correlation functional within the generalized gradient approximation was used [30–32]. The electronic plane-wave cutoff energy was set to 350 eV, and all structures were optimized until the energy differences converged to $<10^{-4}$ eV and the atomic forces fell below 0.01 eV/Å ($1 \text{ Å}=1 \times 10^{-10} \text{ m}$). In detail, an orthorhombic Bi_2S_3 unit cell ($a=11.59359 \text{ Å}$, $b=3.99799 \text{ Å}$, $c=11.16062 \text{ Å}$) with k -point meshes of $1 \times 3 \times 1$ was used in the initial calculations (Bi atoms: 8; S atoms: 12). To investigate the impact of iodine incorporation, the I- Bi_2S_3 structure was modeled using a $\text{Bi}_{13}\text{S}_{18}\text{I}_2$ unit cell with a 6% iodine content. This model featured k -point meshes of $1 \times 1 \times 2$ and a supercell with a doubled z -axis to account for disordered Bi atoms (Bi atoms: 26; S atoms: 36; I atoms: 4). The convergence conditions were consistent with those used for Bi_2S_3 optimization. The energy band structure of the optimized cells was calculated, and the total DOS (TDOS) and the projected DOS (PDOS) were analyzed to investigate the electronic structural variations induced by iodine doping.

2.3 Preparation and performance testing of scaffolds

PLLA/I- Bi_2S_3 scaffolds were fabricated using a custom-built selective laser sintering (SLS) system. The mass ratio of PLLA to I- Bi_2S_3 powder was 98:2. The sintering parameters were as follows: a scan speed of 100 mm/s at a laser power of 0.6 W and a scan speed of 200 mm/s at a laser power of 0.8 W. A layer thickness of 0.1 mm was maintained during fabrication. Scaffolds measuring $15 \text{ mm} \times 5 \text{ mm} \times 1.5 \text{ mm}$ were fabricated for the mechanical and electrical tests, whereas smaller scaffolds ($\Phi 4 \text{ mm} \times 1.5 \text{ mm}$) were used in the biological experiments. The electrical conductivity of the scaffolds was tested using a three-electrode system, confirming their suitability for conductive applications. The photothermal performance of the scaffolds, including time–temperature curves and photothermal stability, was investigated under 1064-nm laser irradiation at 1.5 W/cm^2 .

2.4 ROS analysis of scaffolds

To investigate thermally induced active radical generation, electron spin resonance (ESR) testing was conducted using 5,5-dimethyl-1-pyrroline N-oxide (DMPO) as a radical capture agent. The ESR spectra were recorded under 1064-nm irradiation to identify various ROS, including hydroxyl radicals ($\cdot\text{OH}$) and singlet oxygen ($^1\text{O}_2$). DMPO (100 mmol/L)

was used to detect $\cdot\text{OH}$, whereas 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMP, 100 mmol/L) was used to capture $^1\text{O}_2$. To further quantify ROS generation, 1,3-diphenylisobenzofuran (DPBF) was used to detect the $^1\text{O}_2$ levels via UV–Vis spectroscopy. Similarly, 3,3',5,5'-tetramethylbenzidine (TMB) and O-phenylenediamine (OPD) were used to measure the $\cdot\text{OH}$ levels.

2.5 Antitumor performance of scaffolds

2.5.1 Live/Dead staining assays

The MG-63 human osteosarcoma cell line was incubated in a complete cell culture medium containing 10% fetal bovine serum and 1% penicillin–streptomycin. The cells were cultured in Dulbecco's modified Eagle's medium and maintained at 37°C under 5% CO_2 atmosphere and 95% humidity. MG-63 cells were seeded into six-well plates at a density of 5×10^5 per well and incubated overnight under standard culture conditions. Subsequently, the PLLA, PLLA/ Bi_2S_3 , and PLLA/I- Bi_2S_3 scaffolds were added after the cells adhered to the wall. For the laser-treated group, the cells were irradiated with a 1064-nm laser (1.5 W/cm^2) until the temperature reached 45°C , followed by 5 min of irradiation. After an additional 24 h of co-culture, the cells were stained with calcein acetoxymethyl ester/propidium iodide (Calcein-AM/PI) and Hoechst 33342. The staining process involved removing the culture medium and the scaffolds, washing the cells three times with phosphate-buffered saline (PBS), and adding a 500- μL Hoechst 33342 working solution (0.5 μg) containing a 0.5- μL Calcein-AM solution (2 mmol/L) and a 1.5- μL PI solution (1.5 mmol/L). The cells were incubated at 37°C in the dark for 30 min and then visualized using a fluorescence microscope.

2.5.2 Cell counting kit-8 assays and cell adhesion

The MG-63 cells were plated in a 96-well plate at a density of 5×10^3 per well and incubated overnight. Subsequently, the scaffolds were introduced [33]. The cells were irradiated daily with a 1064-nm laser (1.5 W/cm^2) until the temperature reached 45°C , followed by 5 min of irradiation. After 24 h of additional co-culture, cell viability was assessed using a cell counting kit-8 (CCK-8) assay. In addition, the cell adhesion was observed using scanning electron microscopy (SEM) after 5 d of culture.

2.5.3 ROS staining assays

After scaffold addition and co-culture for 24 h, the cells in the laser-treated group were irradiated with a 1064-nm laser (1.5 W/cm^2) until the temperature reached 45°C , followed by 5 min of irradiation. The cells were then stained with 2',7'-dichlorodihydrofluorescein diacetate (DCFH-DA) to

detect ROS. After removing the medium and the scaffolds, the cells were washed three times with PBS, and 500 μL of a DCFH-DA solution (1 $\mu\text{mol/L}$) was added. The cells were incubated at 37 °C in the dark for 30 min and observed using a fluorescence microscope.

2.5.4 HSP assays

Intracellular HSP90 expression was evaluated using Western blotting. MG-63 cells were seeded in six-well plates at a density of 5×10^4 cells/well. After co-culturing the cells and scaffolds, they were subjected to three cycles of heating in a water bath at 37–45 °C, followed by cell lysis. The protein concentration was determined using a protein quantification kit (Beyotime, China). The protein samples were separated by sodium dodecyl sulfate–polyacrylamide gel electrophoresis, transferred to polyvinylidene fluoride membranes, and blocked with 5% bovine serum albumin at 37 °C overnight. The membranes were incubated with primary antibodies overnight and then with secondary antibodies for 1 h at room temperature. The signals were detected using electrochemiluminescence and imaged using a gel imager.

2.5.5 Necrosis and apoptosis detection

To analyze cell death, the MG-63 cells were seeded in six-well plates and co-cultured with the scaffolds. After laser irradiation (as described above), the cells were harvested and washed three times after three days. Annexin V-FITC (5 μL) and PI (10 μL) staining solutions were added to the cells; the mixtures were gently mixed and incubated at room temperature in the dark for 15 min. Finally, the samples were analyzed using flow cytometry to distinguish between the necrotic and apoptotic cells.

2.6 In vivo antitumor experiment of scaffolds

Eighteen female nude mice (4–6 weeks old) were used to establish a subcutaneous osteosarcoma model. After one week of adaptive feeding, human osteosarcoma MG-63 cells were digested with trypsin and resuspended at 1×10^7 cells/mL, and 0.5 mL of the suspension was subcutaneously injected into the dorsal region of each mouse. When the tumor volume reached approximately 100 mm^3 , the animals were randomly divided into three groups, namely, PLLA, PLLA/ Bi_2S_3 , and PLLA/I- Bi_2S_3 , based on the type of scaffold implanted and their exposure to NIR-II irradiation [34].

The surgical procedure involved anesthesia and disinfection, consistent with standard protocols. The tumor mass was exposed by making an incision along its long axis, and bleeding was controlled using a gauze. A cylindrical

scaffold ($\Phi 4 \text{ mm} \times 1.5 \text{ mm}$) was implanted into the center of the tumor tissue, and the incisions were closed layer by layer with sutures. Post-surgical care included disinfection with complex iodine and intramuscular penicillin injections to prevent infection, followed by daily oral penicillin administration for three consecutive days. Two days after surgery, NIR-II laser irradiation (1064 nm, 1.5 W/cm^2) was applied at the tumor site, with the tumor surface temperature monitored in real time using an infrared thermal imaging system (FLIR, Teledyne FLIR Systems, USA). The first irradiation was defined as Day 0, and this timeline was used for all subsequent daily irradiations. The tumor volume was calculated as follows:

$$V_{\text{tumor}} = (L_{\text{tumor}} \times W_{\text{tumor}})^2 / 2 - V_{\text{scaffold}},$$

where V_{tumor} , L_{tumor} , W_{tumor} , and V_{scaffold} represent the tumor volume, tumor length, tumor width, and scaffold volume (19 mm^3), respectively. On Day 14, all mice were sacrificed. The tumor tissues were harvested, fixed in 4% paraformaldehyde, embedded in paraffin, and sectioned for hematoxylin and eosin (H&E) staining. The H&E staining procedure followed standard protocols. In addition, apoptosis in the osteosarcoma tissues was assessed using the terminal deoxynucleotidyl transferase dUTP nick end labeling (TUNEL) assay.

2.7 Statistical analysis

Data are presented as mean $\pm$ standard deviation ($n \geq 3$). The odds ratios and 95% confidence intervals were calculated, and two-tailed tests were conducted to assess statistical significance. A p -value of ≤ 0.05 was considered significant.

3 Results and discussion

3.1 Microstructure analysis of I- Bi_2S_3

The morphology of the I- Bi_2S_3 powder was characterized using TEM (Fig. 1). The I- Bi_2S_3 powder exhibited a typical nanorod-like structure, with an axial length ranging from 0.7 to 1.2 μm based on statistical analysis (Figs. 1a and 1b). According to the EDS elemental mapping analysis, Bi, S, and I were evenly distributed in the entire structure (Fig. 1c), demonstrating the successful synthesis of I- Bi_2S_3 . As shown in Fig. 1d, the crystal structure of I- Bi_2S_3 exhibited the same diffraction peaks as those of Bi_2S_3 (at 24.9°, 28.6°, and 31.8°), corresponding to the (130), (121), and (221) lattice planes, respectively. The two new diffraction peaks at 23.7° and 28.2° correspond to the (130) and (121) planes of the $\text{Bi}_{19}\text{S}_{27}\text{I}_3$ crystal. The selected area electron diffraction (SAED) images of I- Bi_2S_3 (Fig. 1e) further proved that the identical diffraction crystal planes were

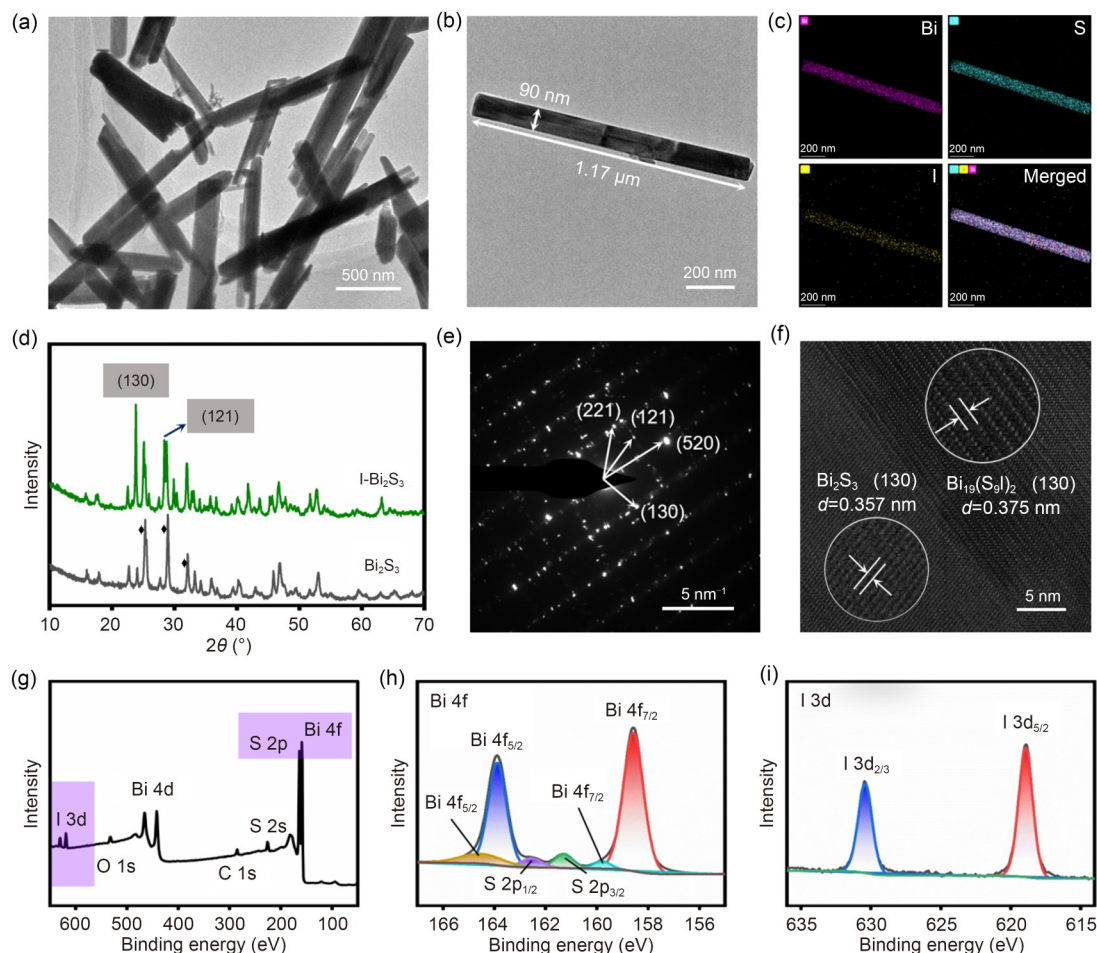


Fig. 1 Microstructure analysis of I-Bi₂S₃. TEM images (a, b) and elemental mappings (c) of I-Bi₂S₃ nanorods. XRD pattern (d), SAED pattern (e), and lattice fringes (f) of I-Bi₂S₃ powder. (g–i) Wide-survey XPS spectrum and Bi 4f and I 3d spectra of I-Bi₂S₃

consistent with those of the (221), (121), and (520) planes of previously reported Bi₁₉S₂₇I₃ crystals [35]. The crystal plane spacing of I-Bi₂S₃ was measured using local Fourier transform. The interplanar spacings of 0.357 and 0.375 nm matched well with the (130) plane of Bi₂S₃ and the (130) crystal plane of Bi₁₃S₁₈I₂, respectively (Fig. 1f). These results demonstrated that iodine doping induced the formation of Bi₁₉S₂₇I₃ and Bi₁₃S₁₈I₂ heterostructures within the I-Bi₂S₃ nanorods.

The XPS spectrum of I-Bi₂S₃ was characterized to analyze the interaction between I atoms and Bi₂S₃. As shown in Fig. 1g, the elemental composition comprised Bi, I, and S, and the chemical states were I 3d, Bi 4d, S 2s, S 2p, and Bi 4f in the entire spectrum. The high-resolution Bi 4f XPS peaks at 158.6 and 163.8 eV were attributed to Bi 4f_{7/2} and Bi 4f_{5/2}, respectively, confirming the formation of Bi–I and Bi–S bonds (Fig. 1h). The different binding energies may account for the existence of bivalent Bi (II) in I-Bi₂S₃ [36]. Furthermore, the I 3d spectrum comprised two peaks at 618.8 and 630.4 eV, corresponding to I 3d_{5/2} and I 3d_{3/2}, respectively [37] (Fig. 1i).

In summary, we successfully synthesized the target I-Bi₂S₃ nanorods through a one-step hydrothermal reaction (Fig. 2a). The successful iodine doping was confirmed by EDS mapping, which also revealed the phase composition and corresponding chemical state of I-Bi₂S₃. Notably, the incorporation of I atoms affected the original Bi₂S₃ crystal structure, resulting in the formation of a new phase. This reconstruction, primarily driven by iodine doping, resulted in a Bi₁₃S₁₈I₂ structure with bivalent Bi (II) [38]. Moreover, to the best of our knowledge, bivalent Bi (II) acts as Bi₂⁴⁺ dimers in I-Bi₂S₃. To verify this hypothesis, the band structure and DOS distribution of I-Bi₂S₃ were investigated using DFT calculations. As shown in Figs. 2b–2d, an optimized Bi₂S₃ unit cell ($a=11.941506$ Å, $b=4.027079$ Å, and $c=11.164037$ Å) was used as an indirect semiconductor, exhibiting a band gap of 1.291 eV. This value is consistent with the optical measurements reported by Groom et al. [39]. For the optimized Bi₁₃S₁₈I₂ unit cell, its lattice constant was $a=15.612$ Å, $b=15.612$ Å, and $c=8.0336$ Å, and the stable atomic coordinates of I atoms in the crystal cell were (0.33333, 0.66667, 0.37498) and (−0.33333,

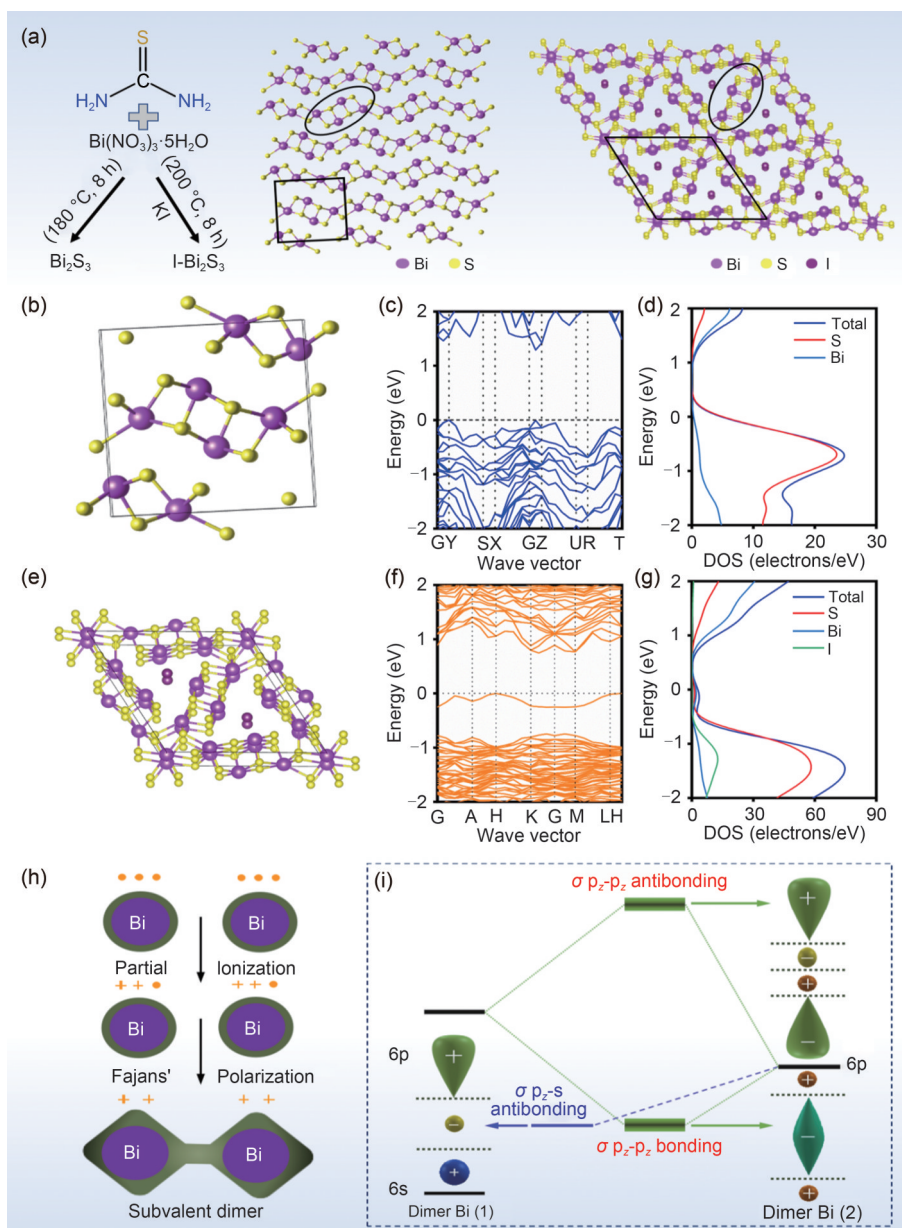


Fig. 2 Overall schematic illustration of the synthesis, crystal/electronic structure evolution, and Bi–Bi bonding mechanism in pristine Bi₂S₃ and iodine-modified I-Bi₂S₃. (a) Synthesis process of Bi₂S₃ and I-Bi₂S₃ (supercell structure diagram). (b, e) Crystal cell models of Bi₂S₃ ($a=11.941506 \text{ \AA}$, $b=4.027079 \text{ \AA}$, $c=11.164037 \text{ \AA}$) and I-Bi₂S₃. (c, d, f, g) Band structure and DOS of Bi₂S₃ and Bi₁₃S₁₈I₂ cells. (h) Formation diagram of bivalent Bi₂⁴⁺. (i) Schematic of partial molecular orbital diagram for Bi–Bi dimer interaction involving 6p_z and 6s orbitals

0.33333, 0.12502). The distance between the two I atoms was 4.0168 Å.

Based on the DFT analysis, a new energy band curve appeared in the energy band structure of Bi₁₃S₁₈I₂ at the top of the VB, resulting in the narrowed bandgap of 0.746 eV. The TDOS and PDOS of Bi₂S₃ and Bi₁₃S₁₈I₂ were calculated to reveal the atomic orbital contributions to the band structure. In Figs. 2c, 2d, 2f, and 2g, the zero-energy level is attributed to the Fermi level. The TDOS curve of the Bi₁₃S₁₈I₂ crystal cell caused by iodine doping exhibits a bend near the Fermi level, which accounts for the

emergence of a new energy level. The new energy level of the VB is dominated by S²⁻ and Bi₂⁴⁺ atoms (Fig. 2g). The formation diagram of bivalent Bi₂⁴⁺ is shown in Fig. 2h. To maintain the overall electroneutrality of the Bi₂S₃ crystal after iodine doping, two electrons from Bi 6p_x and 6p_y orbitals were ionized to compensate for the negative charge of I⁻ ions, resulting in the partial reduction of Bi³⁺ to Bi²⁺. Subsequently, these Bi²⁺ species, located within star-shaped cage-like environments, preferentially formed covalent bonds with adjacent Bi²⁺ ions under Fajans' polarization, resulting in the formation of subvalent Bi₂⁴⁺ dimers. The

stability of the subvalent Bi_2^{4+} dimers in the $\text{Bi}_{13}\text{S}_{18}\text{I}_2$ unit cell can be attributed to the strong interaction between homo-nuclear Bi $6p_z$ orbitals with nearly perfect energy matching. This interaction resulted in the formation of bonding and antibonding molecular orbitals, with the $\text{Bi}(6p_z)\text{--Bi}(6p_z)$ bonding states below the Fermi level stabilizing the Bi–Bi bond (Fig. 2i). Although $\text{Bi}(6s)\text{--Bi}(6p_z)$ antibonding caused slight weakening, the poor energy overlap was insufficient to overcome the overall bond stability. Therefore, the partial occupancy of Bi $6p_z$ orbitals confirms a Bi_2^{4+} subvalent state, supporting the $\text{Bi}_{13}\text{S}_{18}\text{I}_2$ stoichiometry in the P3 space group instead of that in $P6_3/m$, consistent with the XRD results.

Another possible explanation is that iodine doping induced a rearrangement of the initial Bi_2S_3 crystal cells with an orthorhombic space group ($Pnma$), resulting in the formation of a new phase, $\text{Bi}_{19}(\text{S}_9\text{I})_3$, with a hexagonal lattice ($P6_3/m$ space group). Specifically, in its primitive unit cell, three chain-like (Bi_4S_6) subunits are connected by two I atoms and two-thirds of a Bi atom, thereby forming a $\text{Bi}_{2/3}(\text{Bi}_4\text{S}_6)_3\text{I}_2$ unit with a narrow bandgap (0.85 eV): $\text{Bi}(\text{Bi}_2\text{S}_3)_9\text{I}_3$. In this case, as an indirect semiconductor, $\text{Bi}(\text{Bi}_2\text{S}_3)_9\text{I}_3$ can generate electron–hole pairs under NIR-II light excitation. Subsequently, the electrons and holes were relaxed to the band edges, ultimately releasing heat energy in the form of phonons (nonradiative decay process) [40]. In this study, the $\text{Bi}_{19}(\text{S}_9\text{I})_3$ crystal structure was identified via the lattice fringes of the I- Bi_2S_3 powder (Fig. 1f). Therefore, the absorption wavelength of $(\text{Bi}_2\text{S}_3)_9\text{I}_3$ could be extended to the NIR-II region, thereby enhancing the photothermal effect via phonon scattering [41].

3.2 Electrical properties of I- Bi_2S_3 nanorods

Temperature variations can induce the movement of charge carriers (electrons and holes) in pyroelectric materials, directly converting thermal energy into electrical energy. To this end, the electrochemical properties of the I- Bi_2S_3 nanorods were analyzed using an electrochemical workstation and a UV-2700 spectrophotometer. The Mott–Schottky analysis revealed that the I- Bi_2S_3 nanorods are n-type semiconductors, as evidenced by the positive slope of the curve. The CB potential was determined to be -0.35 eV versus the standard calomel electrode (Fig. 3a). As shown in Figs. 3b and 3c, the bandgap widths of Bi_2S_3 and I- Bi_2S_3 were determined from their UV diffuse reflection absorption curves to be 1.23 and 0.88 eV, respectively, using the Tauc plot method [42]. The VB potentials of Bi_2S_3 and I- Bi_2S_3 were determined to be 0.88 and 0.53 eV, respectively, using the equation $E_{\text{VB}} = E_{\text{CB}} + E_g$ (Fig. 3d). These results reveal that I- Bi_2S_3 exhibits a relatively narrow bandgap, which is favorable for electron transition.

The charge transfer resistance of the samples was evaluated using EIS. The Nyquist plot of the I- Bi_2S_3 nanorod exhibited a markedly smaller semicircle radius than that of the Bi_2S_3 nanorod under 1064-nm laser irradiation (Fig. 3e), indicating its lower charge transfer resistance under the same temperature fluctuations [43]. To verify the influence of the material's optical properties on thermoelectric performance, the pyroelectric currents of Bi_2S_3 and I- Bi_2S_3 under 1064-nm laser radiation were measured (Fig. 3f). The I- Bi_2S_3 sample generated a stable average pyroelectric current of approximately 5 μA , which was significantly higher than that of the

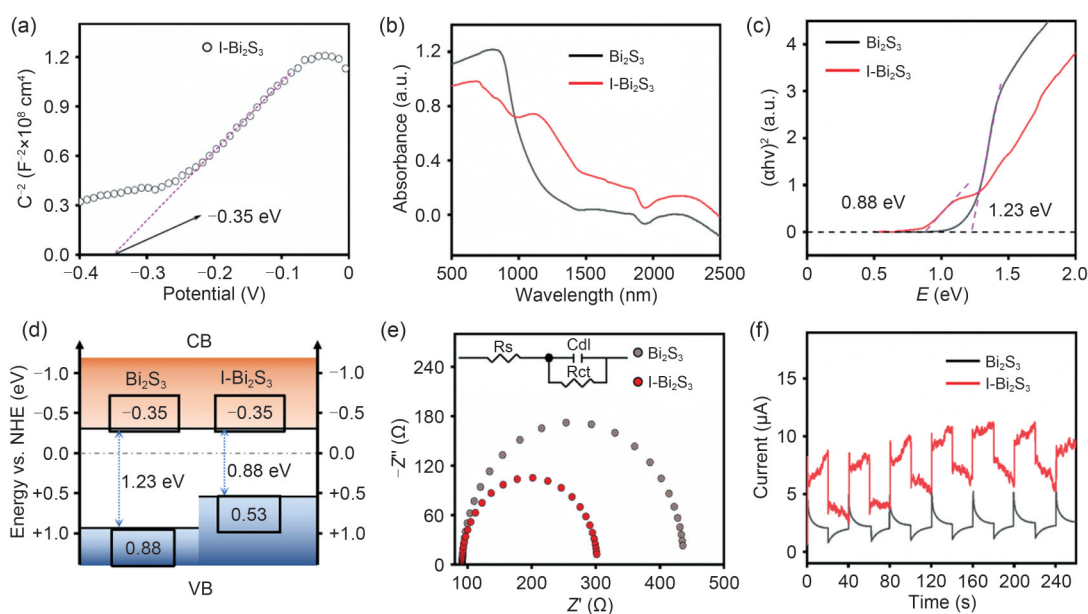


Fig. 3 Electrochemical analysis of I- Bi_2S_3 . (a) Mott–Schottky curves of I- Bi_2S_3 nanorods. (b, c) UV–Vis/NIR diffuse reflection and bandgap of Bi_2S_3 and I- Bi_2S_3 nanorods. (d) Band structure distribution of Bi_2S_3 to I- Bi_2S_3 . (e) EIS Nyquist plots of I- Bi_2S_3 nanorod treatment in the dark and under 1064-nm laser irradiation. (f) Pyroelectric current of Bi_2S_3 and I- Bi_2S_3 nanorods under 1064-nm laser-on/off irradiation

Bi_2S_3 sample. More importantly, the pyroelectric current for I- Bi_2S_3 exhibited good stability, whereas that of Bi_2S_3 only responded instantaneously and quickly disappeared. The results suggest that the electrons of I- Bi_2S_3 can be continuously excited to transition from the VB to the CB under energy input provided by 1064-nm laser irradiation [44]. In contrast, Bi_2S_3 with a wider bandgap requires a shorter-wavelength (higher-energy) light source to achieve sufficient excitation; thus, it fails to sustain a stable current. The difference in current (Fig. 3f) and absorption ability at different wavelength bands (Fig. 3b) confirms that the incorporation of iodine atoms reduces the bandgap of Bi_2S_3 , thereby causing a red shift in the excitation wavelength [45].

In addition, the thermal conductivity (k) of the I- Bi_2S_3 nanorods at temperatures of 30 °C and 50 °C was investigated (Table 1). The thermal conductivity of I- Bi_2S_3 was significantly lower than that of pristine Bi_2S_3 across the measured temperature range. Specifically, the k value of I- Bi_2S_3 decreased to 0.041 W/(m·K) at 50 °C compared with the value of 0.044 W/(m·K) of Bi_2S_3 . This pronounced reduction in k was primarily attributed to the dual low-energy

vibration effect of I- Bi_2S_3 , resulting from the localized vibrational modes introduced by iodine atoms and their interaction with Bi_2^{4+} atoms. The relatively heavy mass and weak bonding of iodine resulted in low-frequency rattling-like vibrations, which were coupled with the intrinsic low-energy modes of Bi_2^{4+} sites. This coupling intensified phonon scattering and disrupted heat transport through the lattice, thereby reducing the thermal conductivity.

Table 1 Thermal conductivity of samples

Temperature (°C)	Thermal conductivity k (W/(m·K))	
	Bi_2S_3	I- Bi_2S_3
30	0.043	0.040
50	0.044	0.041

3.3 Thermoelectric conversion ability of PLLA/I- Bi_2S_3 scaffold

This study aimed to prepare a tissue scaffold with excellent thermoelectric performance for tissue regeneration after tumor resection. As shown in Figs. 4a and 4b, the scaffold

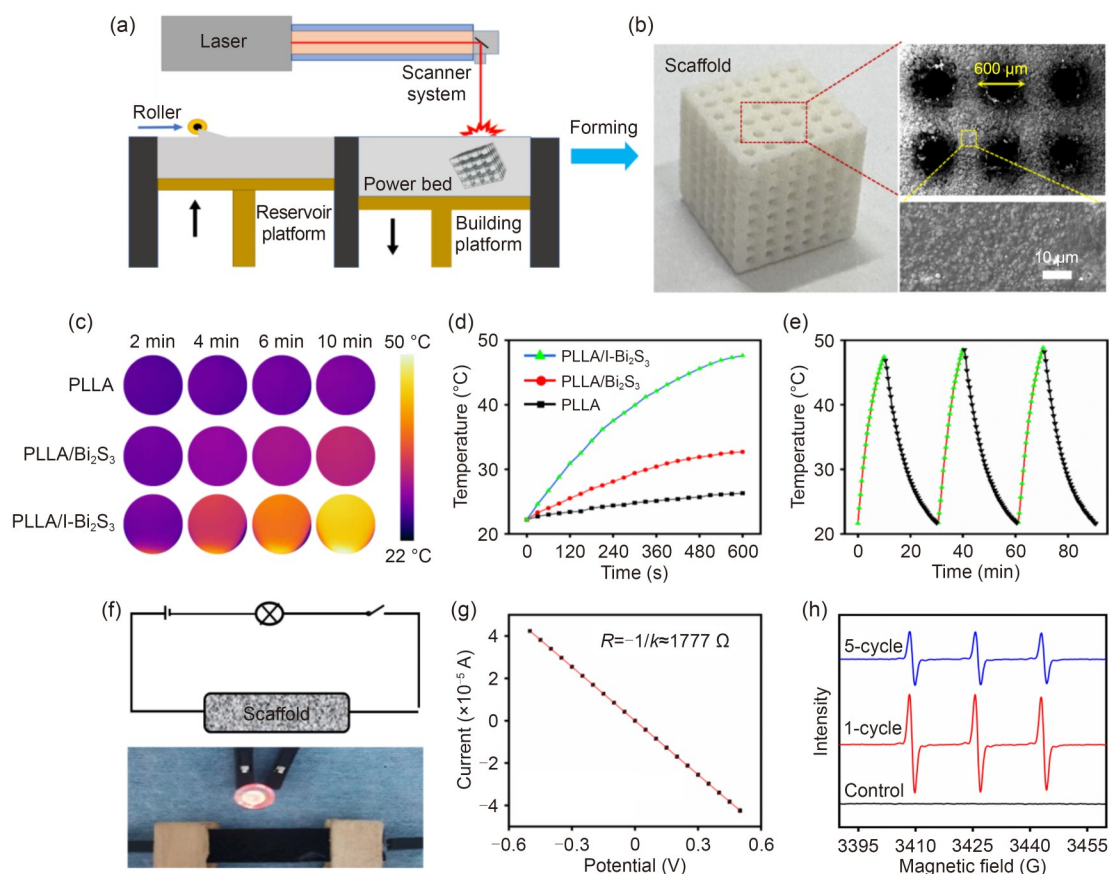


Fig. 4 Photothermal and thermoelectric performance of the scaffold. Preparation process (a) and microstructure (b) of the stent. Thermal imaging images (c) and temperature variation curves (d) of scaffolds under 1064-nm laser illumination. (e) Heating/Cooling curve of PLLA/I- Bi_2S_3 scaffold for three laser-on/off cycles. (f) Conductivity test of PLLA/I- Bi_2S_3 scaffold. (g) Resistance tests of the fitted PLLA/I- Bi_2S_3 scaffold. (h) ESR spectra demonstrating hot-electron generation of PLLA/I- Bi_2S_3 scaffold after one and five laser-on/off cycles

prepared by SLS technology possessed a complicated outline and a fine porous structure with an average pore size of 600 μm . In general, pore sizes of 200–800 μm promote cell adhesion, nutrient exchange, and vascularization, creating an ideal environment for tissue regeneration [46, 47]. The photothermal performance of all scaffolds was evaluated under 1064-nm laser irradiation, and the corresponding time–temperature curves were recorded. Based on the infrared thermography images, the immersion temperatures of all scaffolds, except for the PLLA group, increased with the extension of the illumination time (Fig. 4c). Notably, the immersion temperature of the PLLA/I-Bi₂S₃ group was significantly higher than that of the PLLA/Bi₂S₃ group under the same irradiation time. This result could be because I-Bi₂S₃ achieves stronger absorption and conversion at a wavelength of 1064 nm (Fig. 4d). Moreover, the temperature elevation of the PLLA/I-Bi₂S₃ group exhibited no significant decline after three heating/cooling cycles, demonstrating its good photothermal stability (Fig. 4e). In summary, the PLLA/I-Bi₂S₃ group exhibited relatively good photothermal efficiency compared with the PLLA/Bi₂S₃ group [48].

For thermoelectric scaffolds, good electrical conductivity is an important prerequisite for achieving thermoelectric functions [49]. To visually display the conductivity of the scaffolds, a light-emitting diode (LED) lighting test was conducted (Fig. 4f). The LED indicator successfully lit up after the PLLA/I-Bi₂S₃ scaffold was integrated into the series circuit, indicating that the I-Bi₂S₃ nanorod in the PLLA matrix formed a conductive network for current transmission [50]. In our previous work, we found that filled NPs could be distributed in the grain boundary of the polymer matrix due to the near-zero shearing action of the laser sintering process, thereby resulting in the formation of a connected network in the thermoelectric scaffold. Therefore, although the resistance of the PLLA/I-Bi₂S₃ scaffold reached 1777 Ω (Fig. 4g), it also achieved an internal conductivity function. The hot electrons released by the thermoelectric scaffold were detected using ESR under 1064-nm laser irradiation (Fig. 4h). Compared with the control group, the PLLA/I-Bi₂S₃ group exhibited obvious DMPO peaks after one cooling/heating cycle, confirming hot-electron generation. Furthermore, the intensities of these peaks significantly decreased after five cycles, indicating that the PLLA/I-Bi₂S₃ scaffold consumed the capture agents by continuously producing hot electrons [51].

Therefore, the thermoelectric conversion function of the thermoelectric scaffold can convert phototherapy heat into hot electrons, which can react with surrounding O₂ to generate ROS [52]. Therefore, the ability of the PLLA/I-Bi₂S₃ scaffold to generate ROS in vitro was examined via ESR. As shown in Figs. 5a and 5b, the PLLA/I-Bi₂S₃ scaffold could generate ¹O₂ and $\cdot\text{OH}$ under NIR-II laser irradiation, as evidenced by the 1:1:1 triplet characteristic signal of

TEMP-¹O₂ and the 1:2:2:1 quadruple characteristic signal of the DMPO/ $\cdot\text{OH}$ adduct. To investigate the positive effects after the introduction of I-Bi₂S₃ into the PLLA scaffold, the difference in ROS generation ability between the PLLA/Bi₂S₃ and PLLA/I-Bi₂S₃ scaffolds was explored using a fluorescence colorimetric method. In general, the absorption of DPBF at 416 nm decreases in the presence of ¹O₂, whereas the absorption of TMB and OPD increases in the presence of $\cdot\text{OH}$. The DPBF peak intensity of the PLLA/I-Bi₂S₃ group was lower than that of the PLLA/Bi₂S₃ group under laser irradiation for 10 min (Fig. 5c), indicating that the incorporation of I-Bi₂S₃ enhanced the ¹O₂ generation ability.

Similarly, the PLLA/I-Bi₂S₃ group exhibited a greater $\cdot\text{OH}$ generation ability within the same laser irradiation duration than the PLLA/Bi₂S₃ group (Figs. 5d and 5e). Moreover, for the PLLA/I-Bi₂S₃ scaffold, the peak intensity of the DPBF solution continuously decreased with increasing number of heating/cooling cycles (Fig. 5f). This result indicates that the scaffold exhibits a stable thermoelectric effect under NIR-II laser irradiation [53]. These results confirm that the PLLA/I-Bi₂S₃ scaffold undergoes a light–heat–electric excitation process under NIR-II laser irradiation, ultimately generating desirable ¹O₂ and $\cdot\text{OH}$ radicals (Fig. 5g). In detail, the PLLA/I-Bi₂S₃ scaffold first generates a photothermal effect under NIR-II laser irradiation and then converts the temperature fluctuations of PTT into pyroelectric charges, which ultimately react with surrounding oxygen-containing substances to produce ¹O₂ and $\cdot\text{OH}$. The ROS generation ability of the scaffold primarily originates from the thermoelectric effect, where a temperature gradient effectively drives charge separation and promotes electron transfer, thereby accelerating oxygen reduction. Notably, although photothermal effects can impair mitochondrial respiration and induce ROS generation under substantial hyperthermia, the mild photothermal heating observed in this study (<50 °C) is insufficient to markedly enhance ROS production.

3.4 In vitro antitumor properties of scaffolds

Osteosarcoma MG-63 cells were selected as receptor cells to experimentally evaluate the antitumor efficacy of the scaffolds. The fluorescence live/dead assay showed that the majority of MG-63 cells in the PLLA group were stained green (Fig. 6a), whereas those in the PLLA/Bi₂S₃ and PLLA/I-Bi₂S₃ groups appeared red, indicating that the composite scaffold exhibited good antitumor efficacy under NIR laser irradiation. Notably, the PLLA/I-Bi₂S₃ group exhibited a significantly higher number of dead cells than the PLLA/Bi₂S₃ group, highlighting its superior antitumor efficacy. In contrast, the majority of MG-63 cells in all scaffolds were dyed green in the absence of

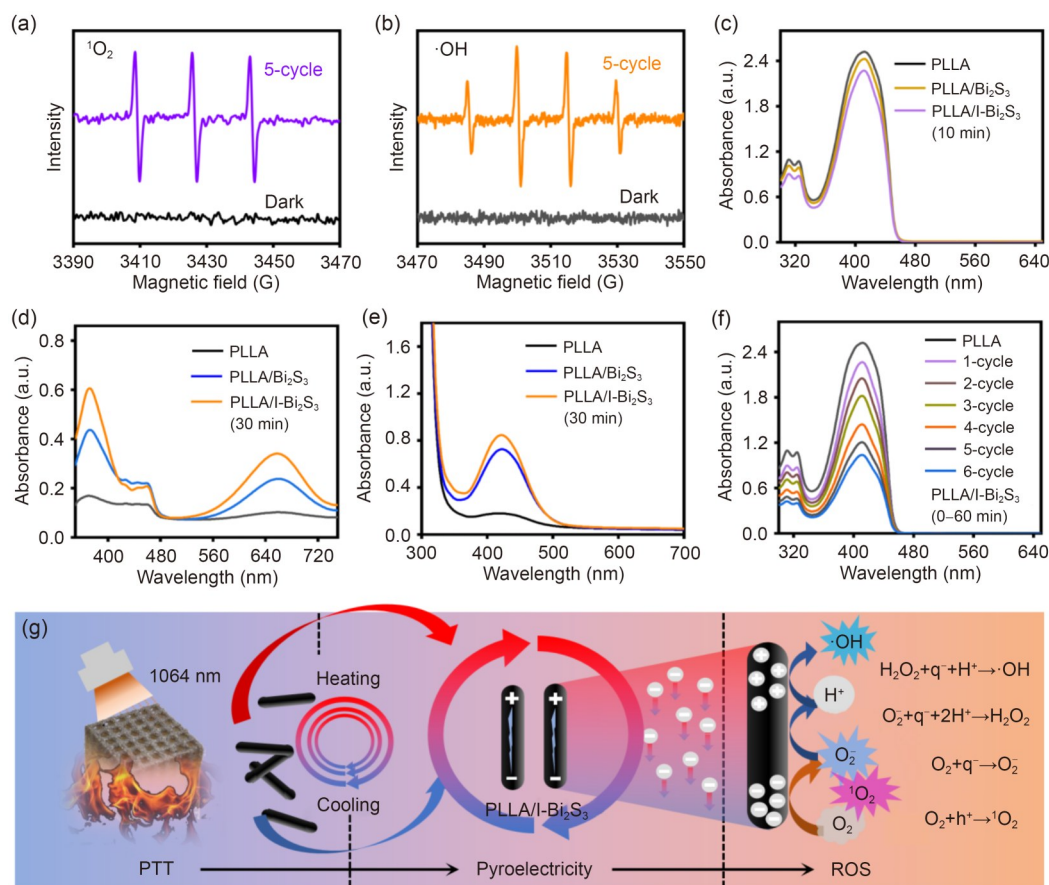


Fig. 5 Evaluation of ROS generation/release from the scaffold. ESR spectra demonstrating $^1\text{O}_2$ (a) and $\cdot\text{OH}$ (b) generation after five laser-on/off cycles. (c) UV-Vis absorption spectra of DPBF after exposure of all scaffolds to 1064-nm laser irradiation. UV-Vis absorption spectra of TMB (d) and OPD (e) after exposure of PLLA/I- Bi_2S_3 scaffold to 1064-nm laser irradiation. (f) UV-Vis absorption spectra of DPBF after exposure of PLLA/I- Bi_2S_3 scaffold to multiple laser-on/off cycles. (g) Mechanism of energy conversion and ROS generation

laser irradiation (Fig. 6b), demonstrating their excellent biocompatibility [54].

Based on the optical density (OD) value changes of the CCK-8 assays (Fig. 6c), the tumor apoptosis rates under NIR laser irradiation were calculated as 64.5% and 84.6% for PLLA/ Bi_2S_3 and PLLA/I- Bi_2S_3 , respectively, whereas the rates without irradiation were only 1.8% and 1.1%, respectively (Fig. 6d). The cell adhesion tests further supported these findings, showing that cells adhered to the PLLA and PLLA/ Bi_2S_3 scaffolds maintained relatively intact morphologies (Fig. 6e), whereas those on the PLLA/I- Bi_2S_3 scaffold were visibly ruptured, emphasizing its enhanced antitumor performance [55]. This conclusion was reinforced by nuclear staining (Fig. 6f), where fragmented nuclei were more obvious in the PLLA/ Bi_2S_3 group than in the other two groups.

Furthermore, the ROS amounts in the tumor cells were examined, confirming that the PLLA/I- Bi_2S_3 scaffold produced more ROS under 1064-nm laser irradiation, as evidenced by the numerous green fluorescence peaks (Fig. 7a) and the increased quantitative ROS values (Fig. 7b). In

contrast, in the absence of laser irradiation, the ROS amounts of all scaffolds were identical [56]. In general, HSPs in tumors are key factors that hinder the efficacy of PTT. Western blotting was used to assess HSP90 expression in MG-63 cells. The results demonstrated that the HSP90 expression in the PLLA group significantly increased after 1064-nm laser irradiation compared with that in the control group without laser exposure. The results demonstrated that HSP90 was rapidly activated in the heat-stress environment. The HSP90 expression in the PLLA/ Bi_2S_3 scaffold was similar to that in the control group. In contrast, the HSP90 expression in the PLLA/I- Bi_2S_3 scaffold was significantly lower than that in the PLLA group, indicating that ROS produced by PLLA/I- Bi_2S_3 effectively suppressed HSP90 [57].

Tumor cell apoptosis and necrosis were examined using flow cytometry. The PLLA group exhibited relatively low apoptosis (Q3) and necrosis (Q2) levels, whereas the PLLA/ Bi_2S_3 and PLLA/I- Bi_2S_3 groups exhibited markedly elevated apoptosis and necrosis levels under NIR-II laser irradiation (Fig. 7d). Based on the quantitative cell apoptosis

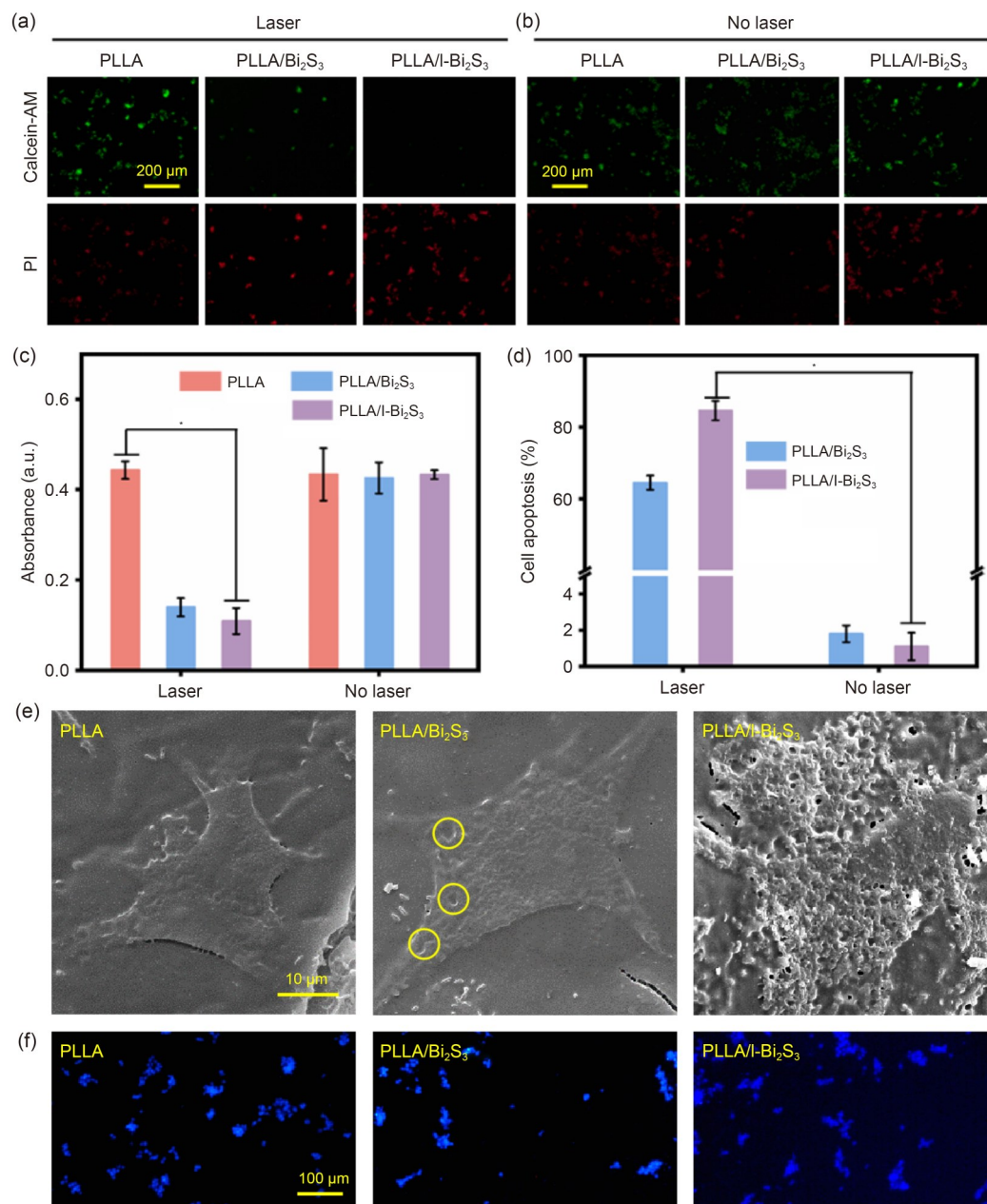


Fig. 6 In vitro antitumor evaluation of the scaffold. Fluorescence images of MG-63 cells under NIR-II laser (a) and without laser (b) exposure. CCK-8 results (c) and corresponding apoptosis rate (d). (e) SEM images of MG-63 cell adhesion. (f) Fluorescence images of cell nuclei in different scaffolds. The error bars indicate mean $\pm$ standard deviation ($n=5$). * $p\leq 0.05$

data (Fig. 7e), the PLLA/I-Bi₂S₃ scaffold exhibited excellent antitumor efficacy compared with the other two scaffold groups [58]. The antitumor mechanism of the scaffold is illustrated in Fig. 7f. After implantation, the scaffold gradually hydrolyzed and released I-Bi₂S₃ into the targeted lesion site. Upon NIR-II laser irradiation, the I-Bi₂S₃ nanorod generated localized heat via PTT, resulting in the thermal ablation of the tumor cells and increased membrane permeability. Simultaneously, the ROS produced by the thermoelectric effect disrupted HSPs, thereby weakening the tumor defense mechanisms and ultimately causing DNA

damage and protein leakage. This innovative strategy integrates the precise release of I-Bi₂S₃ and ROS-mediated cell destruction and photothermal-induced ablation, maximizing therapeutic efficacy while minimizing collateral damage to healthy tissues.

3.5 Evaluation of in vivo antitumor ability of scaffolds

To further verify the antitumor performance of the scaffolds, in vivo animal experiments were conducted using a

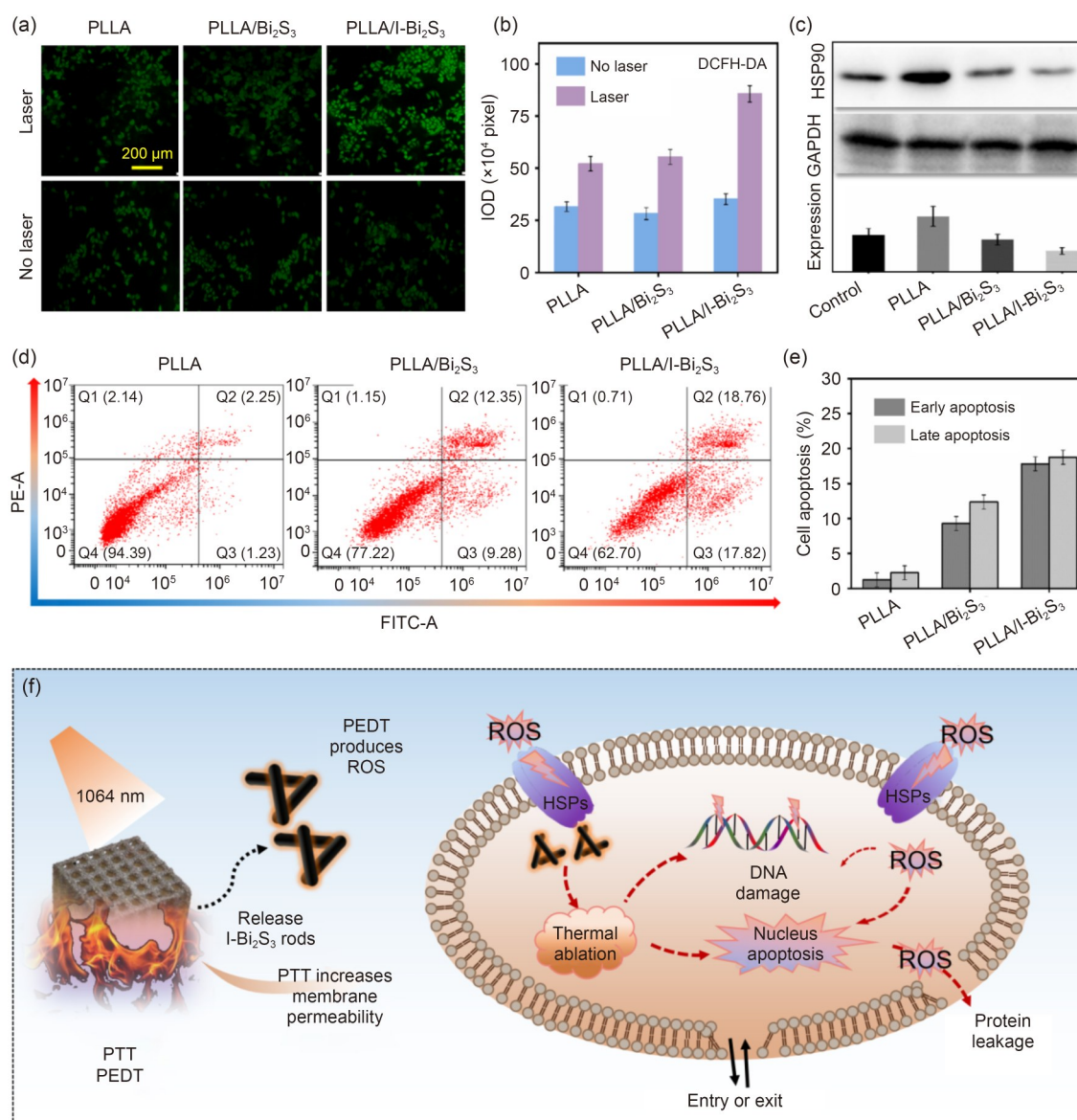


Fig. 7 Mechanistic analysis of scaffold antitumor activity. (a) Fluorescence microscopy images of ROS production. (b) Integrated OD value of fluorescence images of DCFH-DA. (c) Representative Western blotting and quantitative measurement of HSP90 expression. Flow cytometry (d) and relative statistics of tumor cell apoptosis (e). (f) Antitumor mechanism of the scaffold. The error bars indicate mean $\pm$ standard deviation ($n=5$)

nude mouse model of orthotopic osteosarcoma (Fig. 8a). Different scaffolds were implanted into the dorsal region of the nude mice, followed by exposure to NIR-II laser irradiation (1064 nm, 1.5 W/cm², 9 min). According to the infrared thermal images, the temperature of the tumor sites for the PLLA/I-Bi₂S₃ group rapidly increased under NIR-II laser irradiation, reaching 42 °C in 3 min (Fig. 8b). In contrast, the PLLA/Bi₂S₃ group required a longer irradiation time to achieve the same temperature, which was consistent with the results of the in vitro photothermal experiments. Tumor volume changes were recorded after two weeks of cultivation (Fig. 8c). The tumor volume in the PLLA group increased over time, whereas the growth in the PLLA/Bi₂S₃

group was slower, indicating a slight tumor inhibition. In comparison, the tumor volume of the PLLA/I-Bi₂S₃ group was progressively reduced over time, indicating its superior antitumor efficacy.

Fourteen days after surgery, the nude mice were euthanized, and tumor tissues were collected from each group for further evaluation of necrosis and apoptosis using section staining techniques. The photothermal photographs of the tumors excised from the wound sites are shown in Fig. 8d. The tumor size in the PLLA/I-Bi₂S₃ group was significantly smaller than that in the other two groups, demonstrating its superior efficacy in eliminating tumor cells. Tumor necrosis and apoptosis were assessed using H&E staining and

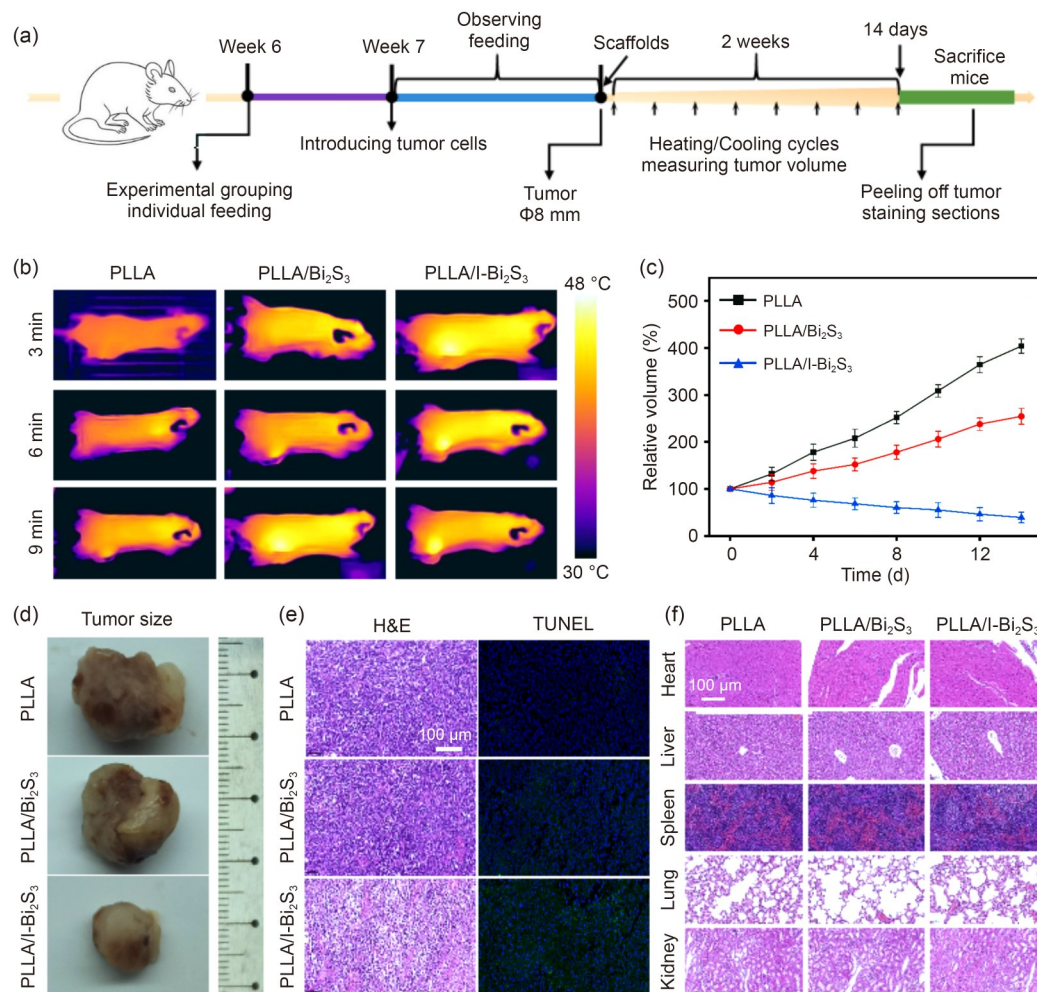


Fig. 8 In vivo antitumor assay of scaffolds. (a) Experimental procedure for tumor treatment in nude mice. (b) Photothermal imaging of nude mice under 1064-nm laser irradiation for 9 min at 1.5 W/cm². (c) Tumor volume changes over time. (d) Weight of isolated tumors for different scaffolds. (e) Representative images of H&E and TUNEL staining in tumor tissues. (f) H&E staining of pathological sections from major organs (heart, liver, spleen, lung, and kidney) of nude mice treated with scaffolds. The error bars indicate mean±standard deviation ($n=3$)

TUNEL staining, respectively. The H&E staining results revealed no significant differences in nuclear staining between the PLLA/Bi₂S₃ and PLLA groups (Fig. 8e). In contrast, the PLLA/I-Bi₂S₃ group exhibited a markedly reduced presence of cells with blue/purple staining (nuclei), indicating its higher level of necrotic osteosarcoma cells.

Similarly, the TUNEL staining results revealed a larger number and more extensive areas of green-stained apoptotic tumor cells in the PLLA/I-Bi₂S₃ group than in the other two groups. These results confirm the superior capacity of the PLLA/I-Bi₂S₃ group to induce apoptosis in tumor cells [59]. Moreover, the in vivo biocompatibility of the scaffolds was evaluated by staining the key organs (heart, liver, spleen, lungs, and kidneys) of each nude mouse (Fig. 8f) [60]. The results revealed no significant inflammatory or physiological damage to any organ, indicating that all scaffolds exhibited good biocompatibility and biosafety.

4 Conclusions

In this study, a novel antitumor PLLA/I-Bi₂S₃ scaffold was synthesized based on photothermal–thermoelectric therapy, providing a promising therapeutic solution for tissue regeneration following osteosarcoma resection. The key findings are as follows:

(1) Bi₂S₃ nanorods with excellent photothermal–thermoelectric functions were synthesized using a hydrothermal reaction method. Iodine doping significantly enhanced the thermoelectric efficiency while broadening the light absorption range to the NIR-II region, thereby enabling effective deep tissue therapy.

(2) Synthesized I-Bi₂S₃ nanorods were integrated into the PLLA matrix using SLS technology, aiming to fabricate a dual-function antitumor scaffold. Under NIR-II laser irradiation, the scaffolds achieved a light–heat–electric conversion process, facilitating localized tumor ablation via photothermal

effects and ROS generation via thermoelectric effects. The scaffolds exhibited an antitumor rate of 84.6% against osteosarcoma cells, demonstrating their superior antitumor performance.

(3) Upon implantation, the scaffold gradually hydrolyzed, releasing I-Bi₂S₃ NPs precisely at the lesion sites. This controlled release minimized collateral damage to healthy tissues while maximizing therapeutic efficacy. In vivo experiments confirmed that the scaffold caused no adverse effects on major organs (heart, liver, spleen, lungs, and kidneys), ensuring its biocompatibility and safety.

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Author contributions JZ (Jun Zan) and JCZ participated in conceptualization, validation, methodology, formal analysis, investigation, and writing of the original draft. JZ (Jie Zeng) was involved in methodology and validation. QY was involved in cell experiment analysis. HYY was involved in supervision, conceptualization, and funding acquisition. YWY and CJS were involved in conceptualization, funding acquisition, supervision, and writing—review & editing.

Declarations

Conflict of interest CJS is an associate editor of *Bio-Design and Manufacturing* and was not involved in the editorial review or the decision to publish this article. The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Ethical approval All animal experiments were approved by the Research Ethics Committee of the Third Xiangya Hospital of Central South University (Approval No. XMSB-2023-0164).

Data availability The datasets generated and/or analyzed during this study are available from the corresponding authors upon reasonable request.

Use of generative AI tools No generative AI tools were used in the preparation of this manuscript.

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